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# TRIBOLOGICAL PERFORMANCE AT HIGH TEMPERATURES OF ALUMINA COATINGS APPLIED BY PLASMA SPRAYING PROCESS ONTO A REFRACTORY MATERIAL

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#### Abstract

Two different Al<sub>2</sub>O<sub>3</sub> coatings manufactured by plasma spraying process were tested under dry sliding contact conditions at high temperatures in order to determine their tribological behavior. At higher temperature than 800 °C, wear rate and mechanisms for Al<sub>2</sub>O<sub>3</sub> coatings have not yet been reported. The results obtained were compared with those from both: an electro-melted Al<sub>2</sub>O<sub>3</sub> refractory used as reference and a pressed & sintered silico-aluminous refractory used as substrate, which were tested under the same conditions. The reference was chosen due to the high performance against wear that this material shows, and the substrate was chosen due to the high potential that this material with low properties against wear shows when it is coated. Crystallographic phases, micro-hardness, Young's modulus and fracture toughness were measured before and after the wear tests, obtaining as results that both Al<sub>2</sub>O<sub>3</sub> coatings showed better wear performance than the pressed & sintered silico-aluminous refractory used as substrate, and even than the electro-melted Al<sub>2</sub>O<sub>3</sub> refractory used as reference. Additionally, in the plasma sprayed coatings, as well as, in the electro-melted Al<sub>2</sub>O<sub>3</sub> refractory, the wear mechanism was controlled by a transformation produced by the increasing in temperature from ductile deformation to brittle deformation, for finally returning to ductile deformation again. The high performance of plasma sprayed coatings was mainly due to the high values of toughness, as well as due to the increasing of the  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> levels and therefore, the hardness during the wear tests because of the high temperature, allowing to these ceramic coatings to be a possible way to protect the conventional pressed & sintered silico-aluminous refractories to replace the expensive nonstructural electro-melted ones currently exposed to abrasive wear in glass, cement and other primary industries.

#### Keywords

Plasma sprayed  $Al_2O_3$  coatings, Electro-melted  $Al_2O_3$  refractory, Pressed & sintered silicoaluminous refractory, Sliding contact, Tribology at high temperature.

#### **1. INTRODUCTION**

Atmospheric Plasma Sprayed (APS) alumina coatings show good performance against abrasive, corrosive and erosive wear at extreme conditions of temperature, oxidation and pressure [1-3]. This is the reason why they have been widely used in applications of high exigency such as automotive, aeronautic and primary industry [4-6]. Specifically, alumina coatings are commonly used for wear resistance due to their high hardness at room temperature [7]. However, the hardness of alumina materials decreases at high temperatures as a result of micro-structural slips activated by the temperature and developed with the applied stresses [8], which is more accentuated by the presence of other oxides [9], which in turn could affect the tribological performance of alumina coatings exposed at high temperature.

Glass and cement industries use refractories subjected to extreme conditions of wear (sliding, abrasion, erosion, etc.) at high temperatures (>1000 °C), which require the employment of expensive electro-melted  $Al_2O_3$  refractories [10]. In order to decrease the wear of the materials involved with a lower cost, different solutions have been proposed [5-6, 11-14]. Among these, thermally sprayed coatings could be an alternative solution to protect the cheap pressed & sintered refractories against tribological failures produced at high temperature. Previous studies have indicated that plasma sprayed ceramic coatings have been manufactured onto pressed & sintering refractory substrates, increasing their useful life when they are exposed to corrosive environments at high temperature and increasing the wear resistance of substrates at room temperature [5-6, 14].

The tribological performance of polycrystalline  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> bulk materials has been studied indicating that until  $\approx 800$  °C, the wear is produced by brittle deformation [15-17]. This last one is developed when the speed and the load applied on the material overcome its mechanical resistance, generating detachment of large and sharp particles, which acts as a third body resulting in high wear rate [18-19]. On the other hand, at higher temperatures, a plastic flux is produced in the material and then, little rounded particles are detached from the surface, producing a protective layer, which reduces the wear rate of the material and is known as ductile deformation [17-19].

The tribological performance of these materials could be related to the microstructure, the phases and their mechanical properties, and in plasma sprayed coatings, these properties are usually different to those in bulk materials due to the fast heating and quenching during the manufacturing of coatings.

The goal of this work is to evaluate the tribological behavior at different temperatures of two thermally sprayed Al<sub>2</sub>O<sub>3</sub> coatings, with different structures and phase compositions, comparing the results with those obtained from both, an electro-melted Al<sub>2</sub>O<sub>3</sub> bulk refractory used as reference, and a conventional pressed & sintered silico-aluminous refractory used as substrate, which were tested under the same conditions that the coatings. The first comparison is justified because the literature about wear performance of thermally sprayed Al<sub>2</sub>O<sub>3</sub> coatings for temperatures as high as those used in this work is poor, and therefore, only there is available information about Al<sub>2</sub>O<sub>3</sub> bulk materials. And the second comparison aims to determine the potential application of the Al<sub>2</sub>O<sub>3</sub> coatings as a possible way to protect the cheap pressed & sintered refractories, to replace the expensive electro-

melted ones subjected to extreme abrasive conditions, but without structural requirements, commonly used in high exigency applications such as: glass, cement and other primary industries.

Wear tests were performed using a ball-on-disk test under dry sliding conditions from room temperature until 1000 °C, and changes in crystalline phases, porosity, micro-hardness, Young's modulus and fracture toughness experienced by the materials due to their heating during tribological tests were also evaluated.

#### 2. MATERIALS AND METHODS

A Sulzer-Metco PTF4<sup>TM</sup> plasma torch was used to spray the alumina coatings according to the parameters listed in Table 1. Two different Al<sub>2</sub>O<sub>3</sub> feedstock powders were used, the first one was constituted by submicrometric agglomerated particles, and the second one was constituted by fused & crushed particles. The substrate was a commercial pressed & sintered silico-aluminous refractory brick, which was cut in shape of discs 25 mm in diameter and 7 mm in height. The arithmetic average of the surface roughness (Ra) was 12  $\pm$  1.1 µm for the substrates, and for this reason, it was not necessary to apply a jet of abrasive particles to give them greater roughness. Therefore, only an air jet 6 bar of pressure was applied to remove particles detached from the surface.

The chemical composition and the crystallographic phases of the feedstock powders, as well as of both, the silico-aluminous refractory used as substrate and the electro-melted alumina refractory used as reference were determined using a Thermo Fisher SCIENTIFIC ARL<sup>TM</sup> OPTIM'X Wavelength-Dispersive X-Ray Fluorescence (WD-XRF) spectrometer and a Bruker D8 ADVANCE X-Ray Cu Ka 1 radiation (DRX) Diffractometer respectively, using the Rietveld method for the quantification of the phases. The particle size distribution of feedstock powders was determined with a Horiba PARTICA LA-950V2 Laser Diffraction (LD) equipment. The shape of the particles, the cross sections, the surface of both coatings and the refractories, as well as the wear tracks were analyzed by Scanning Electronic Microscopy (SEM) using a JEOL JSM IT-300 LV equipment. The cross sections and surfaces of the coatings, as well as the refractories were grinded and polished according to the ASTM E1920 standard [20] in order to obtain an arithmetic average roughness (Ra) lower than 0.2  $\mu$ m. The porosity was measured on the cross sections of the coatings and refractories from images taken by SEM according the indications of the ASTM E2109 standard [21] and using Image J software. Mechanical properties before and after the wear tests of all samples were determined from indentations carried out on the polished surface using a Shimadzu HMV-G 20 equipment, according with the ASTM C-1327 [22] and ASTM E-384 [23] standards. The Vickers micro-hardness, the Young's modulus and the fracture toughness from the Anstis model were calculated according to the Eq. (1-3) respectively:

$$HV = 0.0018544 \frac{P_N}{d^2}$$
 Eq. (1)

Where Hv is the Vickers micro-hardness [GPa],  $P_N$  is the applied normal load on the indenter [N] and d is the average length of the two diagonals produced during the indentation [mm].

$$E = \frac{-\alpha H_K}{\left(\frac{b'}{a'} - \frac{b}{a}\right)}$$
Eq. (2)

Where *E* is the Young's modulus [GPa],  $\alpha$  is a constant ( $\alpha = 0.45$ ),  $H_k$  is the Knoop micro-hardness [Pa], a' and b' are the longer and shorter diagonals respectively produced by the indentation [µm], and *a* and *b* are the geometric constants of the indenter (b/a = 1/7.11).

$$K_{IC} = 0.0016 \sqrt{\frac{E}{H} \frac{P_N}{C^{3/2}}}$$
 Eq. (3)

Where  $K_{IC}$  is the fracture toughness [MPa.m<sup>1/2</sup>], *E* is the Young's modulus [GPa], *H* is the Vickers micro-hardness [GPa],  $P_N$  is the applied normal load on the indenter [N] and *C* is the longest radial crack produced during the indentation [mm].

The wear tests were performed at 25, 500, 750 and 1000 °C with a ball-on-disk tribometer, under a dry sliding contact, without eliminating the formed debris. The wear test conditions are shown in Table 2 following some of recommendations of ASTM G-99 standard [24].

Then, the physical features of the wear tracks on the samples and the morphology of the debris produced in the wear tests were analyzed by SEM using the same equipment aforementioned, equipped with an Energy Dispersive Scanning (EDS) with an Oxford Instruments SDD X-MaxN 80 detector, which was used to determine the chemical elemental composition of these debris. Finally, the wear rate was calculated from the profile curves of wear tracks measured on the samples using a Surtronic S125 profilometer, as well the Eq. (4).

$$WR = \frac{A_s}{1000P_N N_c}$$
 Eq. (4)

Where *WR* is the wear rate  $[mm^3/N.m]$ ,  $A_s$  is the wear track cross section area  $[\mu m^2]$ ,  $P_N$  is the applied normal load [N] and  $N_c$  is the total cycle number.

After tribological tests, the porosity, the crystalline phases, the micro-hardness, the Young's modulus and the fracture toughness of all samples (coatings and refractories) were evaluated again with the same equipments, standards and equations aforementioned, in order to compare the values with those obtained before the wear tests. Each measured property was determined before and after the wear tests over 3 samples, 10 times over each sample, guaranteeing statistical reproducibility and repeatability for all measurements respectively.

#### 3. RESULTS AND DISCUSSION

The chemical composition, the particle size distribution and the crystallographic phases of the feedstock powders used to manufacture the coatings are shown in Table 3. The chemical composition results indicate that mainly  $Al_2O_3$  constitute these materials, with particle size distributions between  $\approx 21 \ \mu\text{m}$  and  $\approx 64 \ \mu\text{m}$ . The main phase in the both feedstock powders was  $\alpha$ - $Al_2O_3$  and additionally, in the fused & crushed one a low quantity of  $\gamma$ - $Al_2O_3$  was also detected. The morphological analysis indicated that the agglomerated powder shows rounded particles constituted by small bonded sub-micrometrical particles (Figure 1 a), evidencing that this powder was manufactured by agglomeration processes, while all particles of the fused & crushed powder were irregular with fracture marks in their surface (Figure 1 b), evidencing that this powder was manufactured by fusing and crushing processes.

On the other hand, the chemical analyses carried out to both, the silico-aluminous refractory used as substrate and the electro-melted  $Al_2O_3$  refractory used as reference are shown in Table 4. The silico-aluminous refractory is constituted mainly by SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub>, while in the electro-melted one the Al<sub>2</sub>O<sub>3</sub> is the predominant oxide, which is combined with SiO<sub>2</sub> and ZrO<sub>2</sub>. The surface analysis of the silico-aluminous refractory suggests irregular grains with low sintering degree and therefore, low density and high porosity, 2.16 to 2.26 g/cm<sup>3</sup> and 16 to 20% respectively, (Figure 2 a, Table 4) typical of pressing and sintering processes of manufacturing. While the structure of the electro-melted Al<sub>2</sub>O<sub>3</sub>

refractory shows low porosity (less than 2%) and therefore high density (3.4 g/cm<sup>3</sup>), (Figure 2 b, Table 4) characteristic of electro-melting processes of manufacturing. The surface analysis of as-sprayed Al<sub>2</sub>O<sub>3</sub> coatings (Figures 3 a-b) evidenced the continuous stacking of micrometrical splats typical of thermal sprayed coatings, indicating that these layers seal the pores of the silico-aluminous refractory. Additionally, on the surface coating sprayed from the agglomerated powder feedstock, sub-micrometrical particles were identified into some splats. (Figure 3 a). Furthermore, the cross-section analysis (Figures 3 c-d) allowed to establish that the thickness of both coatings was similar,  $370 \pm 20 \,\mu$ m for the coating sprayed from the agglomerated powder feedstock. Concerning the porosity (Table 5), it is possible to see that the values before and after the wear tests for all materials and all temperatures evaluated are statistically the same (all p-values are > 0.05), evidencing that there were no sintering processes.

The XRD results indicated that in both alumina coatings,  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>,  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> [25-28] and amorphous phase were the predominant phases (Figure 4 a-b, Table 6). In the electromelted Al<sub>2</sub>O<sub>3</sub> reference refractory used as reference,  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> was the main phase (Figure 4 c, Table 6). In the uncoated silico-aluminous refractory used as substrate, mullite and silica phases were prevalent (Figure 4 d, Table 6). It is important to notice that after the spraying process for both Al<sub>2</sub>O<sub>3</sub> coatings, the  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> from the powder feedstock became mainly in  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> in the as-sprayed coatings. This behavior is due to the initial features of feedstock powders (Table 3) and the spraying parameters (Table 1) as it was previously mentioned by other authors [26-28]. From statistical p-values it was established that in Al<sub>2</sub>O<sub>3</sub> coatings the quantity of amorphous phase,  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> and  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> has a significant change in samples tribologically tested at 1000°C (p-values < 0.05). For its part, the mullite and silica phases percentage in the uncoated silico-aluminous refractory change significantly in the specimens tested from 750 °C. Being of all these, more relevant the increasing of  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>, from the decreasing of  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> and amorphous phases produced in alumina coatings tested at 1000 °C.

Concerning the mechanical properties (Table 7), the micro-indentation results allowed establishing that the micro-hardness of the alumina coatings manufactured from the agglomerated powder feedstock is slightly lower than that of the alumina coating manufactured from the fused & crushed powder feedstock. For both coatings, the values of micro-hardness before and after the wear tests performed at 25, 500 and 750°C are statistically the same (p-values are > 0.05). While at 1000°C, there are differences statistically significant (p-value are < 0.05), due to these hardness significantly increased after their heating at 1000 °C during the wear tests (Table 7), which, could be related with the increasing in the  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> level produced (Table 6, Figure 4 a-b) [4, 29-30].

Particularly, for both coatings the quantities of  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> detected are obtained by two ways: 1) Keeping the corundum phase of powder feedstock during thermal spraying process and 2) Promoting the formation of more corundum phase during the wear tests. The first way is explained through the presence of un-melted particles into the coating, which allow keeping the initial features of the alumina powders. Regarding the second way, the high temperature of wear tests (1000°C), the longtime during pre-heating ( $\approx 2$  h), the wear tests ( $\approx 2$  h) and the cooling ( $\approx 2$  h) of the coatings into the furnace allow to produce the thermodynamic conditions that promote the formation of  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> phase [31-32] as it is shown in Table 6 and Figure 4 a-b.

In the same way, the electro-melted Al<sub>2</sub>O<sub>3</sub> refractory used as reference showed that the hardness after wear test at 1000 °C, is slightly higher than the hardness before the test (p-value < 0.05) (Table 7). This result could be attributed to the percentage decrease of m-ZrO<sub>2</sub> and amorphous secondary phases during the heating carried out for tribological test (Figure 4 c, Table 6). The m-ZrO<sub>2</sub> phase is a polymorph of ZrO<sub>2</sub> with low density and in general with low mechanical [33-35] and tribological properties. Regarding the silico-aluminous refractory used as substrate, although the levels of mullite phase increased and the amorphous phase decreased during wear test carried out at 1000 °C (Figure 4 d, Table 6), the hardness and fracture toughness values decreased (Table 7). This could be due to cohesive failures producing in highly porous materials due to their heating.

Some of the Young's modulus and the fracture toughness values measured before and after the wear tests carried out for all materials, presented change statistically significantly, however these may not be relevant to their tribological performance.

The tribological analysis carried out to both  $Al_2O_3$  coatings has evidenced the development of different wear mechanisms as a function of the test temperature. In both samples evaluated at 25 °C (Figure 5 a), were identified friction marks produced by plastic flow of the asperities in their surface due to sliding contact with the alumina ball. While in those tested at 500 °C and 750 °C (Figure 5 b) high levels of particle detachments produced by

the propagation of cracks in the coatings, were observed, as well as fine and rounded particles which join forming small as island-shaped layers, that in some cases plug the wear tracks produced by the detachment. On the other hand, in the both coatings tested at 1000 °C (Figure 5c) a continuous layer was formed on the wear track, consisting of fine particles, on which was observed a plastic flow.

X. Dong et al. [17] previously studied the change in the tribological mechanism of bulk  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> as a function of temperature. They indicated that the wear was low at the beginning of the tests, with temperatures below 200 °C and applying a normal contact stress greater than 260 MPa at 1.4 mm/s. This was due to the effect of the environmental humidity. Whereas, from this temperature and up to 800 °C, the wear suffered by the material was severe due to the fracture of the material in the zone of contact with the counter-body. Finally, from this last temperature, the wear rate had become moderate again due to the formation of a protective layer. The normal stress applied at the beginning of the tribological tests carried out in this work was 210 MPa, however, there is a similarity in the morphological characteristics observed in the wear tracks of the samples tested at 25 °C (plastic flow), at 500 °C and 750 °C (cracks and fractures), as well as, at 1000 °C (formation of a protective layer), with those reported by X. Dong, et al. [17]. Therefore, the wear stages developed in both Al<sub>2</sub>O<sub>3</sub> coatings are similar to those established by this researcher for bulk  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> materials, despite the fact that the stress applied was lower. This can be related to the higher sliding speed of the ball on the coating (10 cm/s) compared to that of the bulk material (1.4 mm/s), as well as the predominant presence of the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> phase in the coatings compared to predominant presence of the  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> phase

in the bulk material. It is important to notice that the hardness of  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> is lower than that of  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> [36].

G. W. Stachowiak, *et al.* [19] have studied the tribological behavior developed in ceramic materials from applied wear conditions (mainly speed and load) by the sliding contact with a counter-body also ceramic. They found that if the material was able to withstand the stress mechanically applied through the counter body, wear by ductile deformation through plastic flow in the track was produced. While in the case of the material did not have sufficient mechanical strength to withstand the stresses, the wear was produced by brittle deformation. In consequence, cracks and an excessive detachment of the particles were produced. In accordance with that established by W. Stachowiak, *et al.* [19], both types of coatings studied exhibited wear by ductile deformation when they have been tested at 25 °C and 1000 °C. But, when they were tested at 500 °C and 750 °C, the tribological behavior is governed by brittle deformation with higher wear rates due to excessive detachment of the particles.

Despite the fact that the sliding speed (0.1 m/s) and the applied load (5 N) were the same for all tests, a transition from ductile to brittle deformation was detected in the samples evaluated at 25 °C and 500 °C. This result is mainly related the gradual decrease in hardness experienced by alumina as a function of the increase in temperature as previously described [8-9], promoting the detachment of particles. This behavior is more severe in the samples tested at 750 °C, plugging the wear track. The gradual increase of the particles detachment with the temperature has favored the formation of a continuous layer which has protected the wear track of the tested samples at 1000 °C, producing in this case, a wear

transition from brittle to ductile deformation, in which it has been again observed plastic flow (Figure 5 c). It has been previously reported that the presence of wear debris in the friction track can reduce the contact stresses on the surface of coatings, modifying the tribological conditions and therefore the wear resistance [37]. Additionally, the increase in  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> detected after the wear tests carried out at 1000 °C (Figure 4a-b, Table 4), may have had a contribution in the wear transition from fragile to ductile deformation, thanks to the fact that this phase is the hardest of the alumina materials.

Concerning the wear track of the electro-melted  $Al_2O_3$  refractory used as reference, it was detected the same change aforementioned from ductile to fragile for returning to ductile deformation again when the temperature increased (Figure 6). A bigger quantity of debris was detected despite the fact that the tests conditions applied were the same than those for the coatings. this could be due to the low fracture toughness of this material (Table 7), increasing the wear rate compared to the coatings. For its part, the pressed & sintered silico-aluminous refractory used as substrate showed wear by brittle deformation in all cases (Figure 7). This is due to the low mechanical properties of this material shown in Table 7.

The results of wear rate calculated from the track profiles produced in each sample are shown in Table 8. For both types of coatings, their wear rate at 25 °C was of the same order of magnitude as that measured at 1000 °C, being lower than that determined at 500 °C and 750 °C, this due to the ductile-fragile-ductile transition in the tribological behavior previously explained. In addition, at the different temperatures used, the coating sprayed from the agglomerated powder showed slightly lower wear rates than those manufactured

from the fused & crushed powder, as well as, than the electro-melted refractory. This could be related to the superior resistance to crack propagation of this material and the detachment of nanometric particles from un-melted agglomerated grains, in accordance with the results of Y. An, *et al.* [38].

Regarding to the electro-melted refractory, although it has a hardness especially higher than that of the alumina coatings, its wear resistance has been lower. This result could be related, on the one hand to the excessive presence of abrasive debris between the surface and the counter-body. These debris have been analyzed by SEM-EDS, evidencing that they are angular and are composed of Al, Si, Zr and O elements from Al<sub>2</sub>O<sub>3</sub>, the SiO<sub>2</sub> and ZrO<sub>2</sub> phases which constitute the refractory, obtaining as a result, a third abrasive body into the system. On the other hand, the presence of SiO<sub>2</sub> and m-ZrO<sub>2</sub> can significantly reduce the high-temperature hardness of alumina-based materials [9, 33-35].

About the pressed & sintered refractory, it showed higher wear rates than all other materials evaluated, which is basically due to the poor mechanical properties of this material. In addition, the values of friction coefficient (Table 8) measured during the high temperature tribological tests show a downward trend with respect to the temperature for all samples, which could be related to the detachment of fine particles, resulting in the formation of layers that can be protective at high temperatures. It is important to mention that the values obtained from the coefficient of friction are classic compared to the literature, between 0.7 and 0.9 for the alumina against alumina test [19].

#### 4. CONCLUSIONS

- In this work, the tribological behavior at high temperature was evaluated for two different structures and phase compositions of Al<sub>2</sub>O<sub>3</sub> coatings manufactured by APS onto a pressed and sintered silico-aluminous refractory used as substrate. The results were compared with those from an electro-melted alumina refractory used as reference and they were correlated with the phase compositions, porosities and mechanical properties measured before and after the wear tests. The tests were performed using a ball-on-disk test under dry sliding conditions from room temperature until 1000 °C.
- From the results and discussion, it was possible to conclude that the tribological behavior of both plasma sprayed Al<sub>2</sub>O<sub>3</sub> coatings was governed by transitions ductile-fragile-ductile similar to those showed by the electro-melted refractory used as reference and those reported by X. Dong, *et al.* [17] for Al<sub>2</sub>O<sub>3</sub> bulk materials.
- The Al<sub>2</sub>O<sub>3</sub> coatings improved the wear performance of the conventional pressed & sintered silico-aluminous refractory used as substrate, even achieving the wear performance at high temperatures of the electro-melted Al<sub>2</sub>O<sub>3</sub> bulk refractory used as reference, despite le fact that the Al<sub>2</sub>O<sub>3</sub> coatings were mainly constituted by γ-Al<sub>2</sub>O<sub>3</sub>, while the electro-melted refractory was mainly constituted by α-Al<sub>2</sub>O<sub>3</sub>, which indicates that the higher fracture toughness in the coatings contribute to their good wear resistance.

The possible replacement of electro-melted refractories by coated pressed and sintered refractories commonly used in glass and cement industries is only recommended in places subjected to extreme abrasive conditions, but without structural requirements like the combustion ports and the entries of raw materials.

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### TABLES

Parameter	Agglomerated Œrlikon – Metco 6103 <sup>TM</sup>	Fused & crushed Œrlikon – Metco 6062 <sup>TM</sup>		
Current intensity [A]	400	650		
Ar-H <sub>2</sub> flow rate [L/min]	45-	-15		
Nozzle internal diameter [mm]	7	7		
Feeder type	Screw ]	Praxair		
Powder flow rate [g/min]	1	8		
Ar carrier gas pressure [bar]	5.	0		
Ar carrier gas flow rate [L/min]	4.	5		
Spray distance [mm]	100			
Sample translation speed [mm/s]	24			
Sample rotation speed [rpm]	124			
Surface substrate roughness (Ra) [µm]	8 - 10			
Cooling air distance [mm]	12			
Surface substrate preheating temperature [°C]	30	00		
Surface substrate preheating passes	18	12		
Projection time [min]	4			
Number of spraying passes	98	96		

Table 1 Diasm vina poromotor

Table 2. Wear test conditions.					
Counter-body material	Alumina				
Counter-body hardness [GPa]	$18.0 \pm 0.5$				
Counter-body diameter [mm]	6				
Normal load [N]	5				
Linear speed of the sample [m/s]	0.1				
Rotation speed of the sample [rpm]	18,8				
Total number of cycles	20 000				
Total distance of tests [m]	628				
Temperature of tests [°C]	25, 500, 750 and 1000				

Tuble 5. Wain physical chemical foldules of the feedstock powders.				
Properties		Agglomerated Œrlikon – Metco 6103 <sup>TM</sup>	Fused & crushed Œrlikon – Metco 6062 <sup>TM</sup>	
	$Al_2O_3$	$99.9\pm0.3$	$97.6 \pm 0.3$	
Chemical	$SiO_2$		$1.0 \pm 0.1$	
composition	Na <sub>2</sub> O		$0.3 \pm 0.0$	
[wt%]	CaO		$0.3 \pm 0.0$	
	Others	Balance	Balance	
Particle size	d <sub>10</sub>	21.1	25.5	
distribution	d <sub>50</sub>	33.3	36.8	
[µm]	d <sub>90</sub>	63.6	54.2	
Phase	$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	$100.0 \pm 2.8$	$98.8 \pm 1.8$	
analysis [wt%]	$\gamma$ -Al <sub>2</sub> O <sub>3</sub>		$1.2 \pm 0.2$	

Table 3. Main physical-chemical features of the feedstock powders.

Tuble 1. Multi physical chemical features of the fendetories used as substrate and as felerence.					
Prop	erties	Silico-aluminous refractory substrate Erecos ER-40 <sup>TM</sup>	Electro-melted alumina refractory RHI AG Monofrax M <sup>TM</sup>		
	SiO <sub>2</sub>	$52.0 \pm 0.1$	$5.9 \pm 0.1$		
Chemical	$Al_2O_3$	$43.5 \pm 0.2$	$83.8 \pm 0.2$		
	$ZrO_2$		$5.1 \pm 0.1$		
[wt%]	Others	Balance	Balance		
	$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	$2.3 \pm 0.2$	$92.2 \pm 8.3$		
	m-ZrO <sub>2</sub>		$5.3 \pm 0.8$		
Phase	c-ZrO <sub>2</sub>		$0.5 \pm 0.1$		
composition	Cristobalite	$17.9 \pm 0.8$			
[wt%]	Mullite	$45.4 \pm 2.4$			
	Sillimanite	$20.8\pm1.5$			
	Amorphous	$13.6\pm3.9$	$2.0 \pm 0.5$		
Apparent p	orosity [%]	16 - 20	≤2		
Apparent der	nsity [g/cm <sup>3</sup> ]	2.16 - 2.26	3.40		

Table 4. Main physical-chemical features of the refractories used as substrate and as reference	ce.
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Material	Wear tests temperature [°C]	Porosity before wear tests [%]	Porosity after wear tests [%]	ANOVA p-value
Al-O- coating	25		$5.4\pm0.8$	1.000
frage the	500	54.00	$5.4 \pm 0.7$	1.000
from the	750	$5.4 \pm 0.8$	$5.3 \pm 0.9$	0.796
agglomerated powder	1000		$5.3 \pm 0.7$	0.770
Al <sub>2</sub> O <sub>3</sub> coating	25		$4.5\pm0.7$	1.000
from the	500		$4.5\pm0.4$	1.000
fused & crushed	750	$4.5 \pm 0.7$	$4.4 \pm 0.5$	0.718
powder	owder 1000		$4.4\pm0.6$	0.736
Electro-melted	25		$0.58\pm0.05$	0.331
Al <sub>2</sub> O <sub>3</sub>	500		$0.57\pm0.05$	0.513
refractory used as	750	$0.55 \pm 0.08$	$0.52\pm0.08$	0.413
reference	1000		$0.56\pm0.05$	0.742
Pressed & sintered	25		$11.2 \pm 1.1$	0.585
silico-aluminous	500		$11.3\pm0.7$	0.675
refractory used as	750	$11.50 \pm 1.30$	$11.1\pm0.9$	0.435
substrate	1000		$11.0\pm1.3$	0.401

Table 5. Porosity measured before ar	nd after the wear tests	for all materials evaluated.
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•		Wear tests	Before	After	
Material	Variable	temperature	wear tests	wear tests	ANUVA n-velue
		[°C]	[wt %]	[wt %]	p-value
		25		$14.81 \pm 2.43$	0.935
	$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	500 750	$14.72\pm2.41$	$14.99 \pm 1.78$ 13.84 + 2.81	0.779
		1000		$20.76 \pm 3.92$	0.001
$Al_2O_3$ coating		25		$70.33 \pm 5.72$	0.956
from the	$\gamma$ -Al <sub>2</sub> O <sub>2</sub>	500	$70.19 \pm 5.45$	$71.32 \pm 6.71$	0.685
agglomerated	12 - 3	1000		$72.88 \pm 7.02$ 68 44 + 4 39	0.353
powder		25		$14.86 \pm 3.82$	0.886
	Amorphous	500	$15.00 \pm 3.22$	$13.69 \pm 4.02$	0.462
	Allorphous	750	15.09 ± 5.22	$13.28 \pm 3.25$	0.228
		1000		$10.80 \pm 3.36$	0.010
		25 500		$5.03 \pm 0.79$ $5.23 \pm 0.58$	0.826
	$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	750	$5.55 \pm 0.81$	$5.23 \pm 0.33$ $5.44 \pm 0.72$	0.752
Al O spating		1000		$15.13 \pm 1.45$	0.000
Al <sub>2</sub> O <sub>3</sub> coating		25		$84.02 \pm 6.65$	0.992
from the	$\gamma$ -Al <sub>2</sub> O <sub>3</sub>	500 750	$84.05\pm6.70$	$84.02 \pm 3.42$ 83.21 ± 5.57	0.990
rused & crushed		1000		$75.13 \pm 6.98$	0.000
powder		25		$10.35 \pm 0.83$	0.894
	Amorphous	500	$10.40 \pm 0.79$	$10.75 \pm 1.11$	0.436
	rinoiphous	750		$11.35 \pm 0.83$	0.020
		25		$9.74 \pm 0.32$ 93.58 + 6.42	0.030
	41.0	500		$93.54 \pm 5.98$	0.692
	$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	750	$93.23 \pm 8.34$	$93.28 \pm 8.12$	0.779
		1000		94.17 ± 7.14	0.584
	m-ZrO <sub>2</sub>	25	$5.68 \pm 0.83$	$5.29 \pm 0.83$	0.307
Electro moltad Al O		300 750		$5.42 \pm 0.29$ 5.55 + 0.58	0.570
Electro-melted $Al_2O_3$		1000		$5.18 \pm 0.44$	0.116
		25		$0.44\pm0.03$	1.000
used as reference	$c-ZrO_2$	500	$0.44\pm0.09$	$0.42 \pm 0.09$	0.625
	2	1000		$0.49 \pm 0.05$ 0.45 + 0.07	0.147
		25		$0.69 \pm 0.09$	0.333
	Amorphous	500	$0.65 \pm 0.09$	$0{,}62\pm0.08$	0.442
	Amorphous	750	0.05 ± 0.09	$0,68 \pm 0.09$	0.466
		1000		$0.20 \pm 0.02$	0.000
		23 500		$2.30 \pm 0.23$ 2 37 + 0 13	0.764
	$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	750	$2.33 \pm 0.21$	$2.33 \pm 0.18$	1.000
		1000		$2.83\pm0.21$	0.000
		25		$17.95 \pm 0.80$	0.787
	Cristobalite	500 750	$18.05\pm0.83$	$18.27 \pm 1.20$ 19.06 ± 1.44	0.051
		1000		$19.45 \pm 1.62$	0.030
		25		$20.76 \pm 1.49$	0.242
Pressed & sintered	Sillimanite	500	$21.61 \pm 1.64$	$18.48 \pm 1.30$	0.000
silico-aluminous		/50		$18,01 \pm 0,51$ $17.03 \pm 1.67$	0.000
refractory used as substrate		25		$45.39 \pm 2.41$	0.985
	Mullito	500	45 37 + 2 12	$46.73 \pm 2.84$	0.243
	Munite	750	45.57 ± 2.12	$47.65 \pm 2.71$	0.051
		1000		$48.12 \pm 2.78$	0.024
		25 500		$0.40 \pm 0.05$	
	Trydimite	750		$0.98 \pm 0.15$	
		1000		$2.40\pm0.75$	
		25		$13.60 \pm 3.92$	0.585
	Amorphous	500 750	$12.64\pm3.80$	$13.75 \pm 1.18$ 11.07 $\pm 2.72$	0.398
	-	1000		$11.97 \pm 2.02$ $10.17 \pm 2.04$	0.093

Table 6.	Crystallo	ographic	phases	detected	before and	after the	wear test	ts for all	materials	evaluated.
		0								

Material	Variable	Wear tests temperature [°C]	Before wear tests	After wear tests	ANOVA p-value
		25		$10.04 \pm 0.39$	0.884
		500		$10.08\pm0.38$	0.657
	Hardness [GPa]	750	$10.02 \pm 0.17$	$10.15\pm0.34$	0.299
		1000		$12.72\pm0.35$	0.000
Al <sub>2</sub> O <sub>3</sub> coating		25		$245.0 \pm 11.8$	0.367
spraved from the	Young's modulus	500		$246.0\pm9.1$	0.419
agglomerated	[GPa]	750	$249.4 \pm 9.3$	$253.5 \pm 12.5$	0.418
powder		1000		$251.7 \pm 7.4$	0.549
L		25		$3.76\pm0.18$	0.894
	Toughness	500	0.55 0.15	$3.78\pm0.14$	0.650
	$[MPa.m^{1/2}]$	750	$3.75 \pm 0.15$	$3.77\pm0.16$	0.777
		1000		$3.75\pm0.19$	1.000
		25		$10.38\pm0.12$	0.860
		500		$10.32\pm0.14$	0.419
	Hardness [GPa]	750	$10.37 \pm 0.13$	$10.40\pm0.23$	0.725
		1000		$12.26\pm0.24$	0.000
$Al_2O_3$ coating		25		$272.9\pm20.1$	0.887
from the	Young's Modulus	500	274.2 22.2	$278.8 \pm 21.0$	0.656
fused & crushed	[GPa]	750	$2/4.3 \pm 23.3$	$278.9 \pm 18.7$	0.633
powder		1000		$274.4 \pm 18.1$	0.992
P · · · · · · ·	Toughness [MPa.m <sup>1/2</sup> ]	25		$3.71\pm0.33$	0.945
		500	$3.72\pm0.31$	$3.74\pm0.26$	0.878
		750		$3.75 \pm 0.23$	0.809
		1000		$3.47\pm0.29$	0.040
		25		$13.18\pm0.43$	0.359
		500	$13.02\pm0.32$	$13.08\pm0.40$	0.716
	Hardness [GPa]	750		$12.98\pm0.15$	0.727
		1000		$13.31\pm0.17$	0.025
Electro-melted		25		$213.7\pm30.6$	0.952
Al <sub>2</sub> O <sub>2</sub> refractory	Young's Modulus	500	2145 252	$226.4\pm26.0$	0.332
used as reference	[GPa]	750	$214.5 \pm 27.3$	$221.0\pm35.1$	0.650
		1000		$219.3\pm35.2$	0.738
		25		$2.55\pm0.40$	0.664
	Toughness	500	0.60.000	$2.58\pm0.36$	0.797
	$[MPa.m^{1/2}]$	750	$2.62 \pm 0.30$	$2.53\pm0.35$	0.545
		1000		$2.53 \pm 0.32$	0.525
		25		$4.06 \pm 0.41$	0.929
		500		$4.04\pm0.51$	0.871
	Hardness [GPa]	750	$4.08 \pm 0.57$	$4.07\pm0.35$	0.963
		1000		$3.48\pm0.32$	0.012
		25		$25.7\pm4.6$	0.099
Pressed & sintered	Young's modulus	500	20.0 2.5	$26.1\pm4.0$	0.114
silico-aluminous	[GPa]	750	$28.9 \pm 3.5$	$25.2\pm2.4$	0.015
refractory used as		1000		$26.8\pm2.9$	0.162
substrate		25		$1.48\pm0.20$	0.908
	Toughness	500		$1.37\pm0.27$	0.345
	$[MPa.m^{1/2}]$	750	$1.47 \pm 0.18$	$1.35\pm0.29$	0.284
		1000		$1.15\pm0.26$	0.006

Table 7. Mechanical properties measured before and after the wear tests for all materials evaluated.

Material	Wear tests temperature [°C]	Wear rate [mm <sup>3</sup> /N.m]	Friction coefficient	Wear mechanism
$Al_2O_3$ coating	25	$2.73 \times 10^{-5} \pm 9.74 \times 10^{-6}$	$0.80\pm0.10$	Ductile deformation
sprayed from the	500	$2.90 \times 10^{-4} \pm 4.48 \times 10^{-5}$	$0.80\pm0.05$	Brittle deformation
agglomerated	750	$1.03 \times 10^{-3} \pm 1.96 \times 10^{-4}$	$0.80\pm0.03$	Brittle deformation
powder	1000	$2.32{\times}10^{\text{-5}}\pm8.49{\times}10^{\text{-6}}$	$0.75\pm0.04$	Ductile deformation
$Al_2O_3$ coating	25	$3.33 \times 10^{-5} \pm 7.11 \times 10^{-6}$	$0.97\pm0.01$	Ductile deformation
sprayed from the	500	$6.45 \times 10^{-4} \pm 1.48 \times 10^{-4}$	$0.92\pm0.02$	Brittle deformation
fused & crushed	750	$1.25 \times 10^{-3} \pm 3.18 \times 10^{-4}$	$0.88\pm0.02$	Brittle deformation
powder	1000	$3.50 {\times} 10^{\text{-5}} \pm 4.37 {\times} 10^{\text{-6}}$	$0.72\pm0.03$	Ductile deformation
	25	$2.55 \times 10^{-4} \pm 8.30 \times 10^{-5}$	$0.85\pm0.05$	Ductile deformation
Electro-melted $Al_2O_3$	500	$1.75 \times 10^{-3} \pm 2.32 \times 10^{-4}$	$0.83\pm0.03$	Brittle deformation
refractory used as	750	$3.90 \times 10^{-4} \pm 5.88 \times 10^{-5}$	$0.77\pm0.03$	Brittle deformation
reference	1000	$6.00 \times 10^{-4} \pm 2.34 \times 10^{-5}$	$0.74\pm0.04$	Ductile deformation
Pressed & sintered	25	$1.07 \times 10^{-2} \pm 6.61 \times 10^{-3}$	$0.80\pm0.10$	Brittle deformation
silico-aluminous	500	$1.59 \times 10^{-2} \pm 1.19 \times 10^{-3}$	$0.84\pm0.04$	Brittle deformation
refractory used as	750	$2.41{\times}10^{\text{-2}} \pm 9.05{\times}10^{\text{-3}}$	$0.75\pm0.06$	Brittle deformation
substrate	1000	$2.31{\times}10^{\text{-2}} \pm 9.28{\times}10^{\text{-3}}$	$0.70\pm0.10$	Brittle deformation

Table 8. Wear rate and friction coefficient for all materials evaluated.

#### LIST OF FIGURE CAPTIONS

Figure 1. Morphology of feedstock powders: a) Agglomerated and b) Fused & crushed.

Figure 2. Surface structure of: a) Pressed & sintered silico-aluminous refractory used as substrate and b) Electro-melted Al<sub>2</sub>O<sub>3</sub> refractory used as reference.

Figure 3. As-sprayed structures obtained: a-b) Surface of the  $Al_2O_3$  coatings manufactured from the agglomerated and the fused & crushed feedstock powders respectively. c-d) Cross-section of the  $Al_2O_3$  coatings manufactured from the agglomerated and the fused & crushed feedstock powders respectively.

Figure 4. XRD patterns at 25, 500, 750 and 1000°C of: a)  $Al_2O_3$  coating from the agglomerated powder, b)  $Al_2O_3$  coating from the fused & crushed powder, c) Electrometed  $Al_2O_3$  refractory used as reference and d) Pressed & sintered silico-aluminous refractory used as substrate.

Figure 5. Characteristics identified on the surface of polished  $Al_2O_3$  coatings tribologically tested at: a) 25 °C, b) 500 and 750 °C et c) 1000 °C.

Figure 6. Characteristics identified on the surface of electromelted refractory tribologically tested at: a) 25 °C, b) 500 and 750 °C et c) 1000 °C.

Figure 7. Characteristics identified on the surface of pressed and sintered refractory evaluated from room temperature until 1000 °C.



Figure 1



Figure 2



Figure 3.



Figure 4







Figure 5







Figure 6



Figure 7