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WEAR BEHAVIOR AT HIGH TEMPERATURES OF ZrO₂ - Al₂O₃ PLASMA SPRAYED COATINGS AND AN ELECTRO-MELTED AZS REFRACTORY

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Abstract

Two ZrO₂-Al₂O₃ atmospheric plasma coatings were applied to a pressed and sintered aluminosilicate refractory brick, which were compared to an electro-melted alumina-zirconia-silica (AZS) refractory typically used in the glass industry for its high temperature wear resistance. The samples were subjected to sliding contact using a ball-on-disk test at 25, 500, 750, and 1000 °C and then, physical, crystallographic, and mechanical characterizations were performed before and after the wear tests in order to monitor the wear behaviour. Both ZrO₂-Al₂O₃ coatings exhibited better wear performance than the electro-melted AZS refractory at all temperatures evaluated, owing to their high hardness and fracture toughness, which were mainly related to the increase of α-Al₂O₃ and t-ZrO₂ phases from γ-Al₂O₃ and the crystallization of amorphous phases respectively. In the same way, both coatings also showed the same wear mechanisms at all temperatures evaluated. At 25 °C, wear by ductile deformation was identified, while at 500 and 750 °C, wear by brittle deformation was produced, and at 1000 °C, the transition of wear from brittle to ductile deformation was evident. For the electro-melted AZS refractory, until 750 °C, wear by brittle deformation was developed, while at 1000 °C, wear by both, brittle and ductile deformation was detected.

Keywords

ZrO₂-Al₂O₃ coatings, AZS refractory, high temperature, tribological behavior.

1. INTRODUCTION

Electro-melted alumina-zirconia-silica (AZS) refractories are widely used in glass melting furnaces [1-4] because of their thermal shock resistance and resistance to both corrosive and tribological wear when they are subjected to molten glass [1]. However, these refractories are the most expensive components within the glass industry. Therefore, reducing the costs associated with installation, operation, and maintenance of these refractories is critical for the glass industry to compete with plastic, aluminum, and other types of packaging.

Alumina is used in AZS refractories because of its high hardness and good corrosion resistance, despite its relatively low fracture toughness compared to zirconia. In contrast, zirconia exhibits high strength and good fracture toughness, but has low hardness [1, 5]. The tetragonal to monoclinic phase transformation produced above 1000 °C in ZrO₂ causes a significant change in volume (3% – 5%), which can lead to cracking during cooling of this material. However, in AZS refractories, this volume change can be dissipated by the presence of a glassy phase promoted by the silica contained in these materials, which constitutes approximately 30% of their volume [1].

Plasma sprayed zirconia-alumina coatings used as thermal barriers have been studied by other authors for their mechanical properties and oxidation resistance [6-7]. These coatings demonstrate good mechanical and chemical properties, as well as good oxidation resistance without compromising their thermal behavior. Furthermore, the combination of the high hardness of alumina with the low thermal conductivity of zirconia contributes to the development of both the thermal shock and wear resistance of these coatings [7]. In addition to the crystalline phases of alumina and zirconia, amorphous phases are also present [8-10],

which could play an important role in the tribological performance of the coatings when heated above the crystallization temperature (920 °C) [8].

In this study, the tribological behavior of three $ZrO_2-Al_2O_3$ based materials at different temperatures was analyzed. Two $ZrO_2-Al_2O_3$ coatings were thermally sprayed on a pressed and sintered aluminosilicate brick, and the properties were compared to an electro-melted AZS refractory. To the best of our knowledge, previous studies have not reported on these materials and conditions. Furthermore, the results suggest that $ZrO_2-Al_2O_3$ coatings could be applied to protect conventionally pressed and sintered refractories subjected to extreme abrasive conditions, thereby replacing expensive electro-melted refractories for non-structural applications.

2. MATERIALS AND METHODS

To manufacture the ceramic coatings, either a $ZrO_2-Al_2O_3$ Eutectic–Castolin 25088TM (agglomerated and sintered) or a $ZrO_2-Al_2O_3-Y_2O_3$ Tosoh TZ-3Y20ATM (mixed and crushed) feedstock powder was thermally sprayed on a substrate using a Sulzer-Metco PTF4TM plasma torch according to the parameters listed in Table 1. The substrate was a conventionally pressed and sintered aluminosilicate refractory brick ER-40TM manufactured by ERECO S.A., which was cut into short samples (25 mm in diameter and 7 mm in height).

Additionally, an electro-melted alumina-zirconia-silica (AZS) refractory Monofrax CS5TM was used as the reference sample.

The chemical and crystallographic characterization of the feedstock powders was carried out by wavelength-dispersive X-ray fluorescence (WD-XRF) and X-ray diffraction (XRD) using

a Thermo Fisher SCIENTIFIC ARL™ OPTIM'X spectrometer and a Bruker D8 ADVANCE X-Ray Cu K α 1 radiation (XRD) diffractometer, respectively. Additionally, the Rietveld method was used to quantify the phases identified by XRD using the following COD cards: 1) α -Al₂O₃ (9009674), 2) γ -Al₂O₃ (1010461), 3) t-ZrO₂ (1525706), 4) m-ZrO₂ (1010912) and 5) Mullite (9001567). Furthermore, the particle size distribution was determined by laser diffraction (LD) with a Horiba PARTICA LA-950V2, and their morphological characterization was performed by scanning electron microscopy (SEM) using a JEOL JSM IT-300 LV microscope.

The cross-section and surface of the three samples (the two coatings and the AZS refractory used as a reference) were ground and polished according to the ASTM E1920 standard [11] with the aim of obtaining an arithmetic average roughness (Ra) lower than 0.2 μ m to evaluate their structure, mechanical properties, and tribological performance. The structures of both the coatings and the electro-melted refractory were evaluated by SEM using the previously mentioned microscope. This evaluation was carried out on both the as-sprayed surface and the polished cross section of the coatings, while only the surface of the electro-melted refractory was analyzed owing to the isotropic characteristics of its structure. The porosity in the structure was measured from images taken at 200 \times by SEM using the backscatter electron beam following the procedure outlined in the ASTM E2109 standard [12] and using Image J software.

The crystallographic phases of all samples were determined by XRD complemented with Rietveld analysis using the same diffractometer, while the mechanical properties were determined from indentations performed on the polished surface using a Shimadzu HMV-G 20 according to ASTM C-1327 [13]. The microhardness and fracture toughness were calculated from residual indents and cracks produced by Vickers indentation, while the

Young's modulus was calculated from the Knoop indentation. It is important to note that in the ASTM C-1327 standard established for Vickers hardness, at least 10 acceptable indentations meeting an acceptance criterion must be obtained by applying a force for 15 s (3 N for the samples evaluated in this study) that does not change the hardness of the material with the applied load. The fracture toughness was calculated from the length of the radial/median cracks (according to $C > 2a$ criterion, where C is the crack length and a is the average of the indent diagonals, measured from the center of the Vickers indent produced by applying a load of 10 N for 15 s as it is shown in [Figure 1 a](#). For the Young's modulus, a load of 3 N was applied to the Knoop indenter as it is shown in [Figure 1 b](#). The micro-hardness, Young's modulus, and fracture toughness were calculated according to [Eqs. \(1\)–\(3\)](#), respectively:

$$H_V = 0.0018544 \frac{P_N}{d^2} \quad \text{Eq. (1)}$$

Where H_V is the Vickers micro-hardness [GPa], P_N is the applied normal load on the indenter [N], and d is the average length of the two diagonals produced during the indentation [mm].

$$E = \frac{-\alpha H_K}{\left(\frac{b'}{a'} - \frac{b}{a}\right)} \quad \text{Eq. (2)}$$

Where E is the Young's modulus [GPa], α is a constant ($\alpha = 0.45$), H_K is the Knoop microhardness [Pa], a' and b' are the longer and shorter diagonals, respectively, produced by the indentation [μm], and a and b are the geometric constants of the indenter ($b/a = 1/7.11$).

$$K_{IC} = 0.0016 \sqrt{\frac{E}{H_v} \frac{P_N}{C^{3/2}}} \quad \text{Eq. (3)}$$

Where K_{IC} is the fracture toughness [MPa.m^{1/2}], E is the Young's modulus [GPa], H_v is the Vickers micro-hardness [GPa], P_N is the applied normal load on the indenter [N], and C is the longest radial crack produced during the indentation [mm] [14].

On the other hand, the wear tests of all samples were carried out using a homemade ball-on-disk tribometer under dry sliding contact at temperatures of 25, 500, 750, and 1000 °C, using an alumina ball 6 mm in diameter as counter-body, applying a load of 5 N during 20000 cycles and rotating the samples at a linear speed of 0.1 ms⁻¹ without eliminating the formed debris, according to the recommendations of the ASTM G-99 standard [15] shown in Table 2.

The wear tracks were analyzed by SEM using the aforementioned microscope as well as by optical microscopy (OM) using an Olympus BX41 microscope. To calculate the wear rate produced for each sample, Eq. (4) was used. The average cross-sectional area (A_s) for each wear track was determined from the analysis of profiles at 10 different points using a Surtronic S125 profilometer as it is shown in Figure 2.

$$WR_{Sample} = \frac{Volume}{Load \times Distance} = \frac{A_s 2\pi r_{wt}}{1000 P_N N_c 2\pi r_{wt}} \quad \text{Eq. (4)}$$

Where WR_{Sample} is the wear rate [mm³/N.m], A_s is the wear track cross-sectional area [μm²], r_{wt} is the wear track radius [mm], P_N is the applied normal load [N], and N_c is the total cycle number. In the same way, to calculate the wear rate produced for each counter-body, Eq. (5) was used. The height (h) for each worn counter-body was determined from 10 different measurements using an electronic Rexbeti micrometer.

$$WR_{Counter-body} = \frac{Volume}{Load \times Distance} = \frac{\frac{1}{3}\pi h^2(3R - h)}{P_N D_T} \quad \text{Eq. (5)}$$

where $WR_{Counter-body}$ is the wear rate [$\text{mm}^3/\text{N.m}$], h is the height of the spheric cap [mm], R is the counter-body diameter [mm], P_N is the applied normal load [N], and D_T is the total distance of the tests [m].

After the tribological tests, the porosity, crystallographic phases, micro-hardness, Young's modulus, and fracture toughness of all samples were evaluated as described above, in order to determine possible changes produced during the wear tests performed at different temperatures. Each property was determined before and after the wear tests using three samples, and the mechanical properties were tested 10 times for each sample, guaranteeing statistical reproducibility and repeatability for all measurements.

3. RESULTS

3.1. Characterization of feedstock powders

The results of the chemical composition, phases, and particle size distribution of the feedstock powders used to manufacture the bimodal and monomodal coatings are summarized in [Table 3](#). Both powders consisted mainly of ZrO_2 and Al_2O_3 , with low quantities of Y_2O_3 , Na_2O , and MgO . Consequently, $m\text{-ZrO}_2$ and $\alpha\text{-Al}_2\text{O}_3$ were identified as crystalline phases. The particle size distribution of the Eutectic–Castolin 25088TM powder was $d_{10}=15.25 \mu\text{m}$ and $d_{90}=94.12 \mu\text{m}$, while that for Tosoh TZ-3Y20ATM was $d_{10}=29.82 \mu\text{m}$ and $d_{90}= 57.12 \mu\text{m}$. Morphological analysis indicated that the Eutectic–Castolin 25088TM powder corresponded to rounded granules composed of micron-sized particles, which are characteristic of those manufactured

by agglomeration and sintering processes (Figure 3 a). For the Tosoh TZ-3Y20ATM powder, sharp particles and fracture patterns were identified, which is characteristic of feedstock processed by crushing (Figure 3 b).

3.2. Structural characterization of samples

Structural analysis of both the as-sprayed surface and the polished cross section of the Eutectic–Castolin 25088TM coating powder was characterized by partially melted particles (Figures 4 a-b) embedded in the splats and lamellae, respectively. The partially melted particles embedded in the molten material produced a bimodal coating on the structure of the substrate. In contrast, the structure of the coating sprayed from the Tosoh TZ-3Y20ATM powder exhibited only splats, lamellae, and pores, which are characteristics of thermally sprayed coatings with a monomodal structure (Figures 4 c-d). The thickness of the bimodal coating manufactured from the Eutectic–Castolin 25088TM powder was $215 \pm 20 \mu\text{m}$, while that of the monomodal coating sprayed from the Tosoh TZ-3Y20ATM was $260 \pm 25 \mu\text{m}$. The porosity value measured on the cross sections of both samples was higher in the bimodal coating ($8.0 \pm 1.0\%$) than in the monomodal coating ($7.1 \pm 1.1\%$) due to the presence of partially melted particles that hindered the stacking of sprayed particles. The structural analysis of the electro-melted AZS refractory revealed zirconia–alumina dendrites in an amorphous matrix phase that was highly densified (Figure 4 e), resulting in a very low porosity ($0.30 \pm 0.05 \%$). Specifically, the white particles in the dendrites shown in Figure 4 e correspond to the tetragonal zirconia phase, while the light gray particles are alpha alumina, and the dark gray region is the amorphous phase. The porosities measured before and after the tribological tests are shown in Figure 5, and the results indicate that the temperatures of the wear tests for all three samples did not promote sintering.

3.3. Crystallographic characterization of samples

The phase analysis established that both coatings consisted mainly of an amorphous phase (> 50 wt.%) with α -Al₂O₃ and γ -Al₂O₃ phases and smaller quantities of t-ZrO₂ and m-ZrO₂ phases (Figures 6 a-b). The γ -Al₂O₃ phase in these coatings was produced from the α -Al₂O₃ phase transformation produced during the thermal spraying process, as previously described [16-22]. These results are in agreement with those of other authors who have reported the presence of large quantities of amorphous phases [23], as well as α -Al₂O₃, γ -Al₂O₃, t-ZrO₂, and m-ZrO₂ phases, in plasma-sprayed coatings from feedstock powders with different proportions of Al₂O₃ and ZrO₂ [23-24]. In comparison, the electro-melted AZS refractory phase analysis indicated that it consisted mainly of amorphous phases (~42 wt.%), t-ZrO₂ (~28 wt.%), α -Al₂O₃ (~22 wt.%), and mullite (~7 wt.%) (Figure 6 c), corresponding to the crystallographic compositions previously reported [25].

Figures 7 a-c show that some phases present in our samples had significant changes after heating to 1000 °C during the tribological tests. In both coatings, the t-ZrO₂ phase increased significantly at the expense of the amorphous phase, as previously reported by other authors [23]. For the electro-melted AZS refractory, a small increase in both phases, t-ZrO₂ and α -Al₂O₃, was produced by crystallization of its amorphous phase. These results indicate that in both coatings, the amorphous phase corresponded to low-crystallinity t-ZrO₂, while the electro-melted AZS refractory consisted of a glassy phase related to the approximately 30 vol% SiO₂ in its composition.

3.4. Mechanical characterization of samples

The results of the mechanical characterization performed before and after the tribological tests are presented in Figures 8 a-c. In general, it is possible to see a slight increase in the hardness

of both coatings after exposure to 1000 °C during the tribological tests, which can be correlated to the changes in the crystalline phases (Figures 6-7 a-c). The Young's modulus for all samples did not indicate significant differences after exposure to different temperatures during the wear tests, which indicates that this mechanical property is not very sensitive to the crystallization of amorphous phases produced at 1000 °C. The fracture toughness did not change for the bimodal coating at any temperature, even at 1000 °C, however, for the monomodal coating, these values were stable until 750°C and decreased at 1000 °C. For the electro-melted AZS refractory, the mechanical properties did not change significantly as a function of temperature, mainly because of the low values of crystallization from the amorphous phase to the t-ZrO₂ and α-Al₂O₃ phases, as well as the minimal changes in porosity.

3.5. Wear characterization of samples

The wear rate results of the ZrO₂-Al₂O₃ coatings and the electro-melted AZS refractory are presented in Figure 9, where it is possible to see that for all materials evaluated, the highest wear rates were obtained at 500 and 750 °C, while the lowest ones were obtained at 25 and 1000 °C, which is related to the crystallographic phases and therefore to the mechanical properties of these materials.

The analysis of wear tracks produced in the three samples showed the development of different tribological mechanisms depending on the test temperature. For both ZrO₂-Al₂O₃ coatings tested at 25 °C, regular wear tracks, with plastic deformation and without any sign of fracture were observed (Figures 10-11 a-c). Meanwhile, at 500 and 750 °C, irregular wear tracks, with signs of microcracks and detachment of particles were evident (Figures 10-11 d-i). Furthermore, at 1000 °C, again regular wear tracks were identified, but this time with continuous layers composed of fine debris, in which plastic flow was observed (Figures 10-11

j-1). In contrast, a regular wear track, with evidence of plastic deformation and cracking mainly produced in the amorphous phase were observed in the wear track of the electro-melted AZS refractory tested at 25 °C (Figures 12 a-c). At 500 °C, in addition to the plastic deformation of the amorphous phase, the detachment of both zirconia and the surrounding amorphous phase was also observed, producing irregular wear tracks (Figures 12 d-f). Additionally, in samples tested at 750 °C, the detachment of both zirconia dendrites and the surrounding amorphous phase was accentuated, producing wear tracks also irregular (Figure 12 g-i). These detachments were even more pronounced in samples evaluated at 1000 °C, despite the discontinuous layer formed from debris (Figure 12 j-1).

The friction coefficient values (Figure 13) measured during the tribological tests did not indicate a particular trend with respect to the temperature. However, in general, they were lower for the samples evaluated at 1000 °C, likely due to the reduction in the severity of the tribological contact by the layer formed from debris. The coefficient of friction values were similar to those previously reported for ZrO₂-based materials against alumina [26].

Figure 14 showed that the counter-body wear rates increased with the temperature until 750°C, and then at 1000°C, these rates decreased, which was probably related to the aforementioned changing in crystallographic phases and mechanical properties, and therefore in the tribological mechanisms. Additionally, from Figures 15, it is clear that on the worn surfaces of the counter-bodies, there were signs of friction traces as expected for abrasive wear, and there is not any sign of wear by adhesion.

4. DISCUSSION

In general, the hardness of all samples is governed by the phases that constitute them. In this case, both ZrO₂-Al₂O₃ based coatings containing ~40 wt.% of alumina phases (α -Al₂O₃ and γ -Al₂O₃) had a higher hardness than the AZS electro-melted refractory whose alumina phase content was ~22 wt.%. Alumina phases are harder than the zirconia and mullite phases present in these three specimens [27-30]. Additionally, the increase in the α -Al₂O₃ phase at the expense of the γ -Al₂O₃ phase when the spray coated samples were heated to 1000 °C was evident and it is widely reported [16-22], while the increase in the α -Al₂O₃ phase detected after heating the electro-melted AZS refractory at 1000 °C was due to crystallization of the amorphous phase. Previous studies have demonstrated crystallization of the amorphous phase in Al₂O₃-ZrO₂ materials from temperatures close to 950 °C [23]. In the same way, the hardness also increased because of the increase in the t-ZrO₂ phase produced by the crystallization of the amorphous phase during the wear tests at 1000 °C. The t-ZrO₂ crystals have higher hardness values (~16 GPa) [31] than those of the ZrO₂ amorphous phase (~12.0 GPa) [31].

Concerning the crack propagation resistance, the AZS electro-melted refractory displayed a lower fracture toughness than both coatings because its amorphous phase corresponded mainly to a glassy phase, whereas the high volume of amorphous phase in the coatings was a low-crystallinity t-ZrO₂. The high propagation of cracks in vitreous materials has been previously reported by other authors [32]. Additionally, the fracture toughness values after the tests at 1000 °C decreased significantly for the monomodal coating, whereas it did not change for the bimodal coating. The fracture toughness of the bimodal coating samples after the high

temperature tribological tests did not decrease because of the presence of partially melted particles that may have contributed to the arrest of the cracks.

The microcracks produced during the sliding of the counter-body on the three samples were due to fatigue fracture, as has been previously reported by other authors for ZrO₂-based materials [33-35]. Fatigue fracture and particle detachment are characteristics of a special kind of abrasive wear mechanism applied only to ceramics called wear by brittle deformation, which is produced when the stresses applied by sliding contact with a hard counter-body are higher than the mechanical resistance of the samples [36]. Additionally, other authors have indicated that the wear resistance of ceramic materials depends on both their hardness and fracture toughness [37-39]. In this case, the Vickers microhardness and fracture toughness of both ZrO₂-Al₂O₃ coatings before and after tribological testing were higher than those of the electro-melted AZS refractory, which makes the latter sample more prone to wear by brittle deformation. Additionally, the adhesion of debris to the wear track surface promotes the formation of layers, as evidenced by samples tested at 1000 °C. The layers modify the conditions of sliding contact, reduce cracks and detachment of particles, and therefore produce a plastic deformation typical of a special kind of abrasive wear mechanism applied only to ceramics called wear by ductile deformation [33, 36, 40]. The microcracks and detachment of particles were more accentuated by increasing temperature when the wear was produced by brittle deformation.

When wear was governed by ductile deformation in samples tested at 25 °C, the wear rates were lower than those tested at 500 and 750 °C, which were controlled by brittle deformation. The increase in wear rate with increasing temperature is because the hardness of the alumina, zirconia, and glass (as the amorphous phase in the electro-melted AZS refractory) decreases with increasing temperature [41-42], followed by a possible softening of the samples as the

temperature of each tribological test increased. This softening may have also affected the alpha alumina ball used as a counter-body, but in a lower proportion owing to its lower porosity (lower than 0.1) than that of the samples. The decrease in the hardness of the samples made them more prone to wear by brittle deformation, increasing the cracking and detachment of particles from the wear track with increasing temperature. However, when the samples were subjected to 1000 °C, some changes in their phases and, perhaps, a partial recovery of their mechanical properties was produced. In both coatings, the amorphous phase crystallized to the t-ZrO₂ phase and γ -Al₂O₃ transformed to the α -Al₂O₃ phase, which increased the content of these phases in the coatings by ~25 wt.% and ~4 wt.%, respectively. On the other hand, in the electro-melted AZS refractory, the crystallization of the amorphous phase to both t-ZrO₂ and α -Al₂O₃ phases was evident, which increased the content of these phases in the sample to ~4 wt.% and ~2 wt.%. These phase transformations may have increased the hardness of the samples, since t-ZrO₂ is harder than the amorphous phase from which it was produced [31] and α -Al₂O₃ is harder than γ -Al₂O₃ [16, 43-44].

The increase in the harder phases in the samples tested at 1000 °C may have helped to counteract the progressive softening produced with increasing temperature, and also may have contributed to the change in the tribological mechanism from wear by brittle deformation to wear by ductile deformation, mainly in the coatings. Additionally, the layer on the 1000 °C samples with apparent plastic flow on the wear tracks indicated that the softening of samples produced the formation of this layer, reducing the severity of the tribological contact. This led to a decrease in the wear rate of the samples evaluated at 1000 °C compared to those tested at 500 and 750 °C.

It is important to note that the wear rate produced at all temperatures was lower for the coatings than for the electro-melted AZS refractory. This is due to the higher hardness and

fracture toughness of the coatings (Figure 8 a), in agreement with other authors who find that the wear resistance of ceramic materials increases with these mechanical properties [37-39]. Additionally, the wear rate produced in the bimodal coating was lower than that in the monomodal coating, suggesting that the presence of partially melted particles, which may have retained some of the agglomerated fine particles in the structure of this coating, can improve its wear resistance.

5. CONCLUSIONS

The tribological behavior at different temperatures of bimodal and monomodal plasma-sprayed $ZrO_2-Al_2O_3$ coatings were evaluated and compared to an electro-melted AZS refractory. The wear behavior of these materials correlated to their mechanical properties, which is a function of their crystalline phases. The mechanical properties and crystalline phases were measured before and after the tribological tests. The changes in the wear mechanism as a function of the temperature were identified.

The wear mechanisms identified in both plasma-sprayed $ZrO_2-Al_2O_3$ coatings were as follows: ductile deformation at 25 °C, brittle deformation at 500 and 750 °C, and ductile deformation at 1000 °C. In contrast, wear by both brittle and ductile deformation was identified at 25 °C for the electro-melted AZS refractory, while wear by brittle deformation was identified at 500 and 750 °C. At 1000 °C, wear owing to both brittle and ductile deformation was identified again. These behaviors were related to possible changes in their mechanical properties owing to the softening produced in the alumina, zirconia, and glass (as the amorphous phase in the electro-melted AZS

refractory) with the increase in temperature, as well as the partial recovery of the hardness by crystallization of harder phases at 1000 °C.

- The ZrO₂-Al₂O₃ coatings produced a higher wear resistance at all evaluated temperatures than the electro-melted AZS refractory, mainly because of their higher hardness and fracture toughness. The hardness of the coatings was higher than that of the AZS refractory, mainly owing to the high content of alumina phases. On the other hand, the fracture toughness of the electro-melted AZS refractory was lower than that of the coatings because the amorphous phase corresponded mainly to a brittle glassy phase, whereas the amorphous phase in the coatings was a low-crystallinity t-ZrO₂.
- The low-cost pressed and sintered refractories coated with a zirconia-alumina plasma sprayed layer could replace the expensive electro-melted AZS refractories for abrasive and non-structural applications, such as combustion ports and batch houses, in the glass and cement industries.

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TABLES

Table 1. Plasma spraying parameters

| Parameter | Agglomerated and sintered | Mixed and crushed |
|---|---------------------------|-------------------|
| Current intensity [A] | 650 | |
| Ar-H ₂ flow rate [L/min] | 45 - 15 | |
| Nozzle internal diameter [mm] | 7 | |
| Feeder type | Screw Praxair | |
| Powder flow rate [g/min] | 16.0 | 22.5 |
| Ar carrier gas pressure [bar] | 5.0 | |
| Ar carrier gas flow rate [L/min] | 4.5 | |
| Spraying distance [mm] | 100 ± 1 | |
| Sample translation speed [mm/s] | 24 | |
| Sample rotation speed [rpm] | 124 | |
| Substrate surface roughness (Ra) [μm] | 8-10 | |
| Cooling air distance [mm] | 12 | |
| Surface substrate preheating temperature [°C] | 300 | |
| Surface substrate preheating passes | 18-26 | |
| Spraying time [min] | 4 | |
| Number of spraying passes | 96 | 92 |

Table 2. Wear tests parameters

| | |
|----------------------------------|-----------------------|
| Counter-body material | Alumina |
| Hardness of counter-body [GPa] | 18.0 ± 0.5 |
| Counter-body diameter [mm] | 6 |
| Normal load [N] | 5.0 |
| Linear speed of the sample [m/s] | 0.1 |
| Total number of cycles | 20000 |
| Radius of tests [mm] | 5 |
| Total distance of tests [m] | 628 |
| Temperature of tests [°C] | 25, 500, 750 and 1000 |
| Heating rate of tests [°C/min]* | 50 |

* The sample and the counter-body were maintained in the furnace of tribometer at the temperature of test during 2 hours before to start each tribological essay.

Table 3. Physicochemical features of the feedstock powders

| Properties | | Agglomerated and sintered | Mixed and crushed |
|-----------------------|----------------------------------|----------------------------------|--------------------------|
| Chemical | ZrO ₂ | 57.52 ± 0.82 | 75.72 ± 0.84 |
| Composition (wt.%) | Al ₂ O ₃ | 38.94 ± 0.43 | 19.70 ± 0.49 |
| | Others | Balance | Balance |
| Phase | m-ZrO ₂ | 60.4 ± 5.0 | 74.4 ± 4.7 |
| Analysis (wt.%) | α-Al ₂ O ₃ | 39.6 ± 4.3 | 25.6 ± 3.9 |
| | Amorphous | ----- | ----- |
| Particle Size | d ₁₀ | 15.25 | 29.82 |
| Distritubtion (μm) | d ₅₀ | 51.15 | 41.82 |
| | d ₉₀ | 94.12 | 57.12 |

LIST OF CAPTIONS

Figure 1. Typical residual indentation to determine: a) Fracture toughness and b) Young's modulus.

Figure 2. Typical wear track cross section to determine the wear rate.

Figure 3. Morphology of the $ZrO_2-Al_2O_3$ based feedstock powders: (a) Agglomerated and sintered and (b) Mixed and crushed.

Figure 4. Microstructures obtained. Bimodal $ZrO_2-Al_2O_3$ coating: a) Surface and b) Cross-section. Monomodal $ZrO_2-Al_2O_3$ coating: c) Surface and d) Cross-section. Electro-melted AZS refractory: e) Surface.

Figure 5. Porosity measured before and after the wear tests for all materials evaluated.

Figure 6. XRD patterns obtained. a) Bimodal $ZrO_2-Al_2O_3$ coating, b) Monomodal $ZrO_2-Al_2O_3$ coating and c) Electro-melted AZS refractory.

Figure 7. Crystallographic phases detected before and after the wear tests. a) Bimodal coating, b) Monomodal coating and c) Electro-melted AZS refractory.

Figure 8. Mechanical properties measured before and after the wear tests. a) Bimodal coating, b) Monomodal coating and c) Electro-melted AZS refractory

Figure 9. Wear rate of all samples evaluated.

Figure 10. Typical wear tracks obtained for the bimodal $ZrO_2-Al_2O_3$ coating tribologically tested at: a-c) 25 °C, d-f) 500 °C, g-i) 750 °C and j-l) 1000 °C.

Figure 11. Typical wear tracks obtained for the monomodal $ZrO_2-Al_2O_3$ coating tribologically tested at: a-c) 25 °C, d-f) 500 °C, g-i) 750 °C and j-l) 1000 °C.

Figure 12. Typical wear tracks obtained for the electro-melted AZS refractory tribologically tested at: a-c) 25 °C, d-f) 500 °C, g-i) 750 °C and j-l) 1000 °C.

Figure 13. Friction coefficient of all samples evaluated.

Figure 14. Wear rate of all counter-bodies evaluated.

Figure 15. Typical wear on the counter-bodies. At 25 °C: a) Bimodal, b) Monomodal, c) Refractory. At 500 °C: d) Bimodal, e) Monomodal, f) Refractory. At 750 °C: g) Bimodal, h) Monomodal, i) Refractory. At 1000 °C: j) Bimodal, k) Monomodal, l) Refractory.

FIGURES

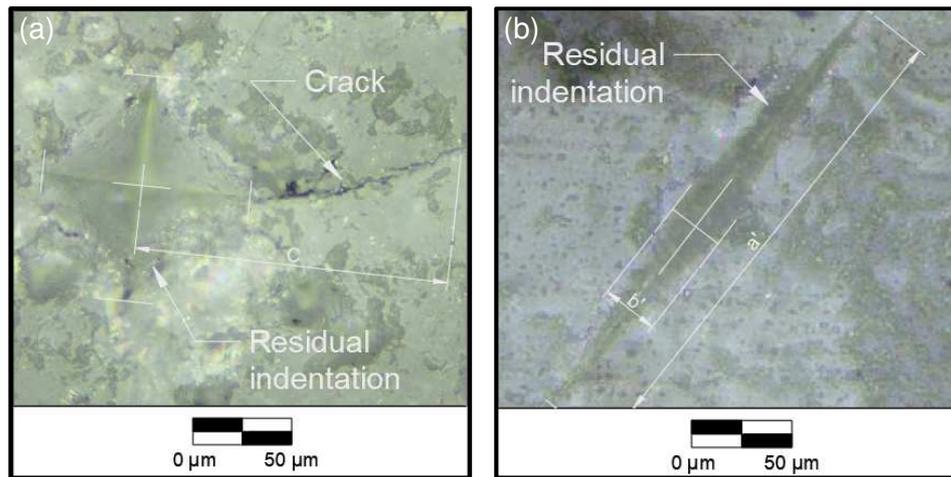


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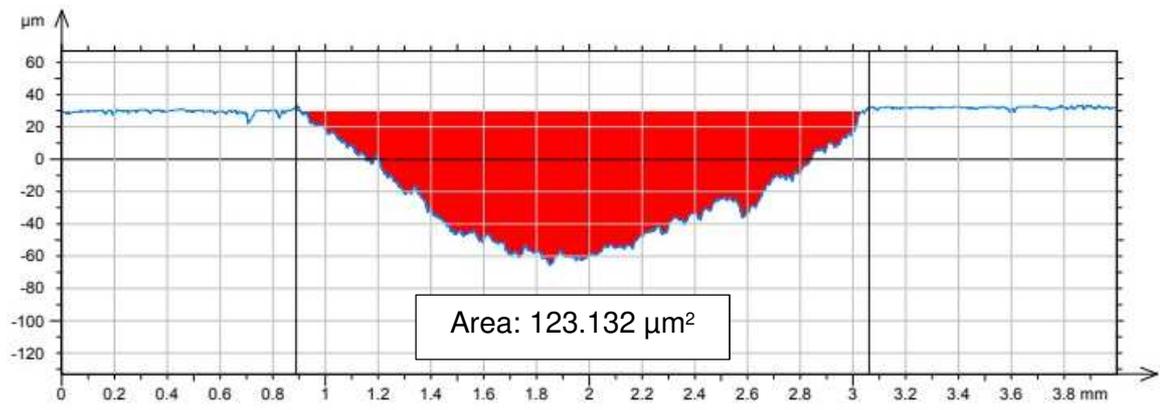


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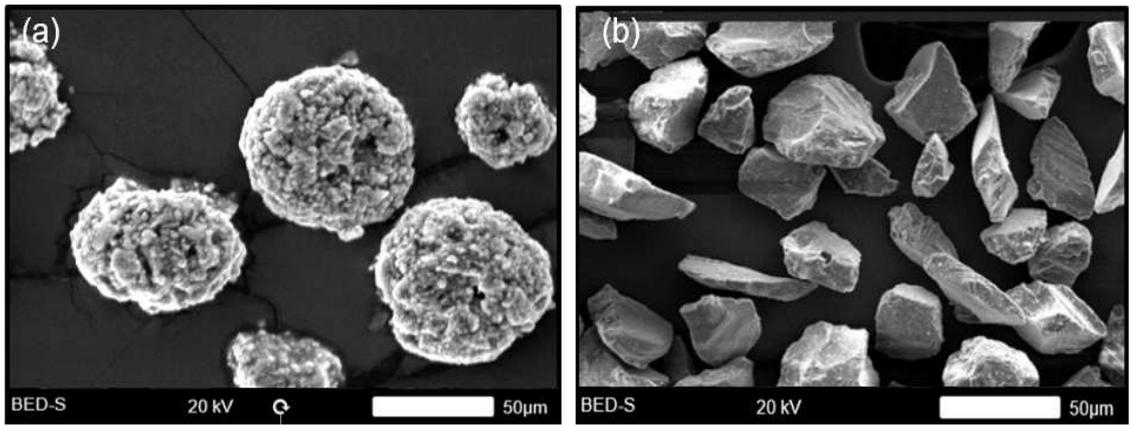


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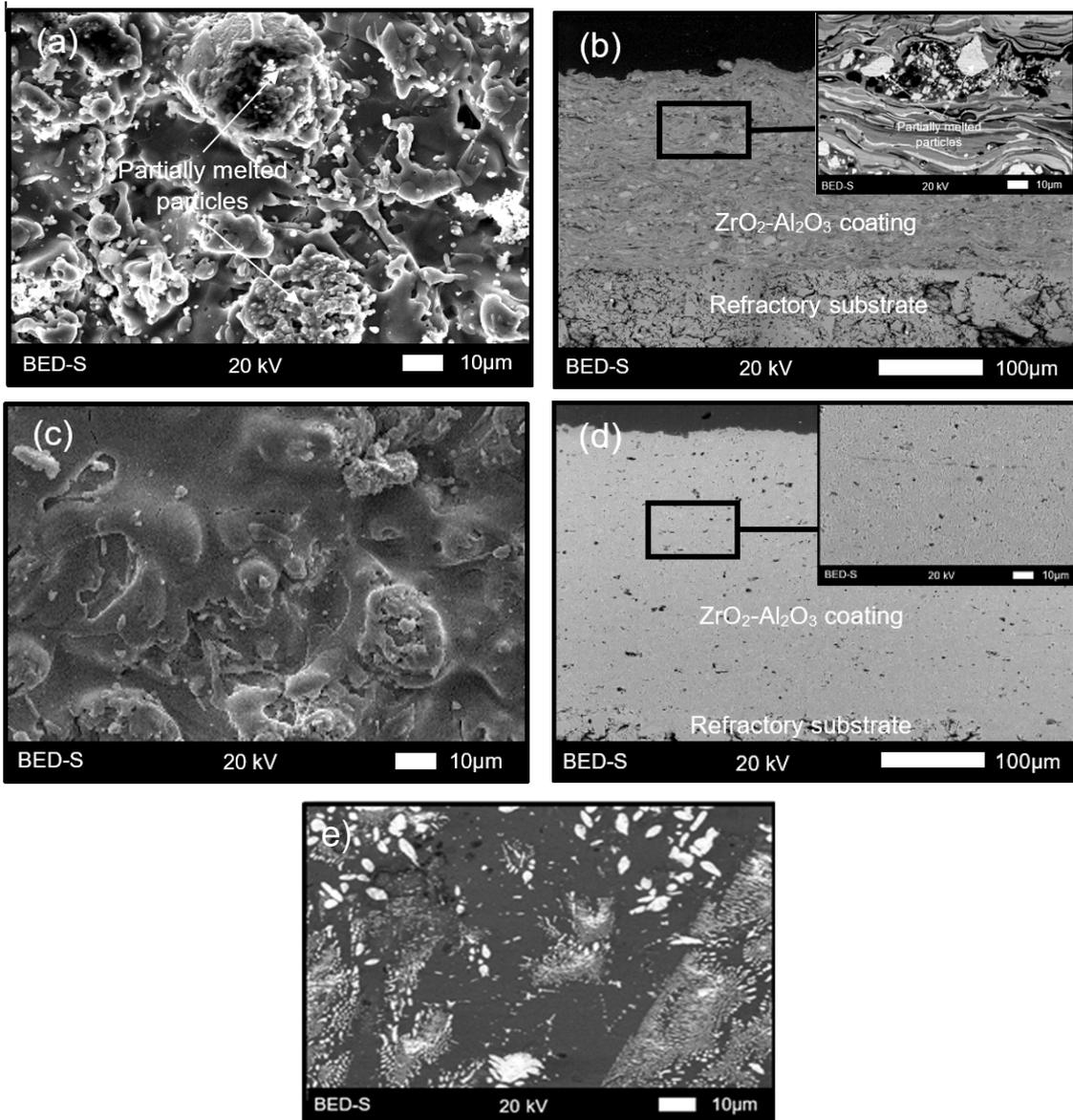


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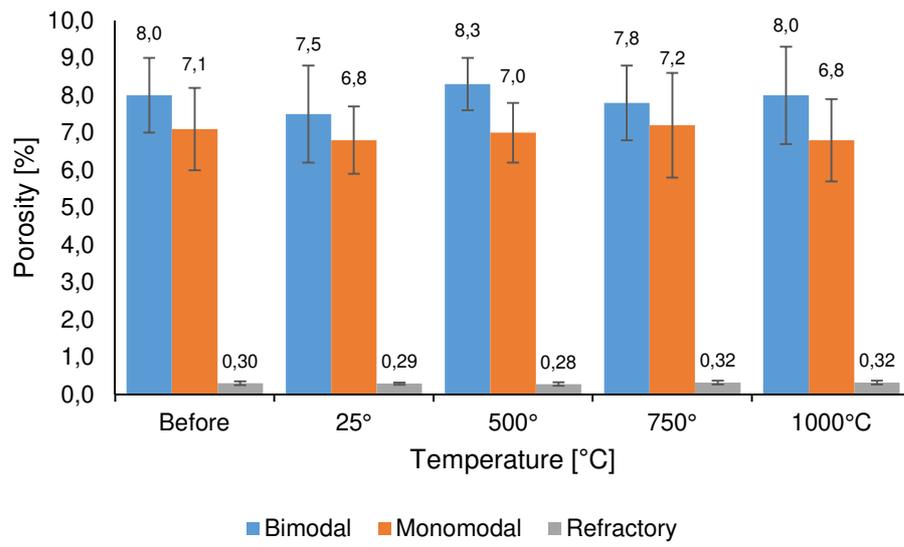
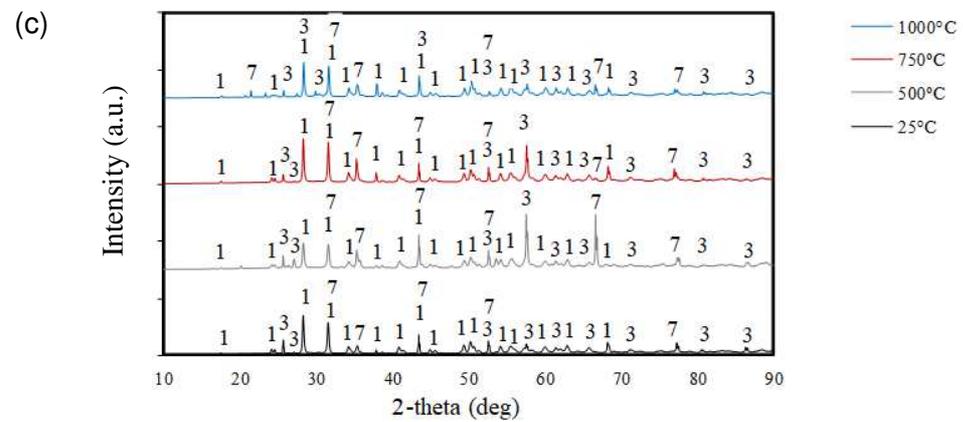
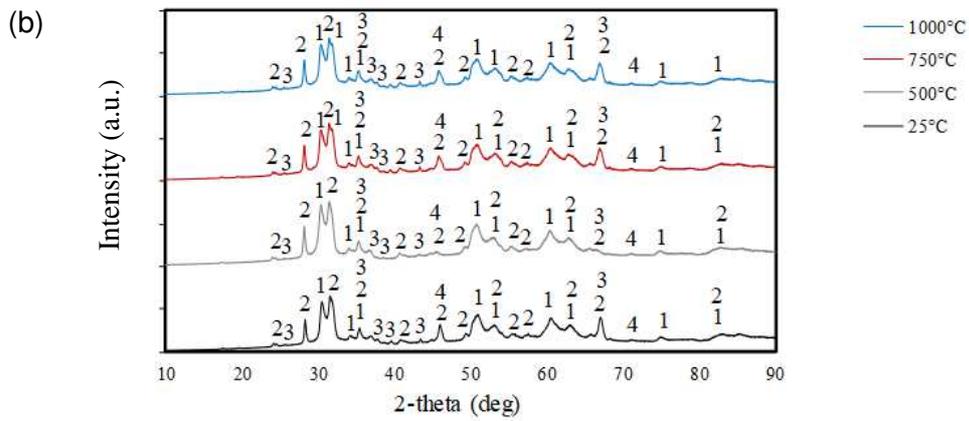
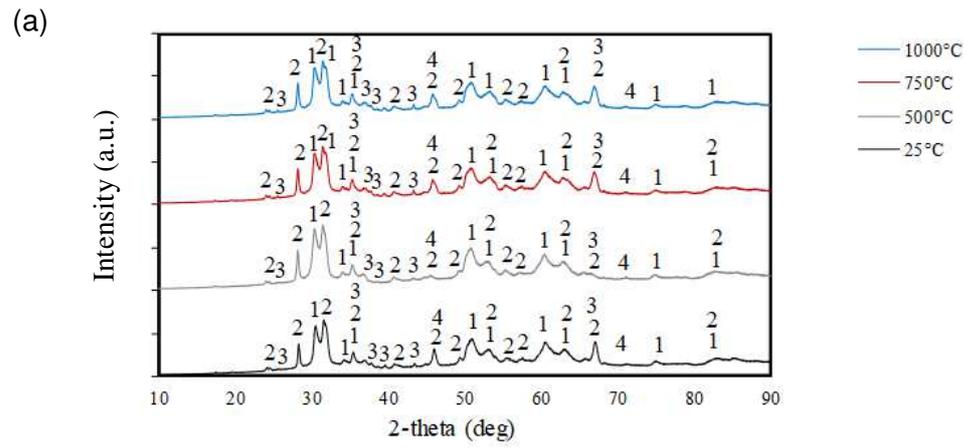


Figure 5.



1. ZrO_2 -t 2. ZrO_2 -m 3. Al_2O_3 - α 4- Al_2O_3 - γ
5. ZrO_2 -c 6. YZr_8O_{14} 7. Mullite

Figure 6.

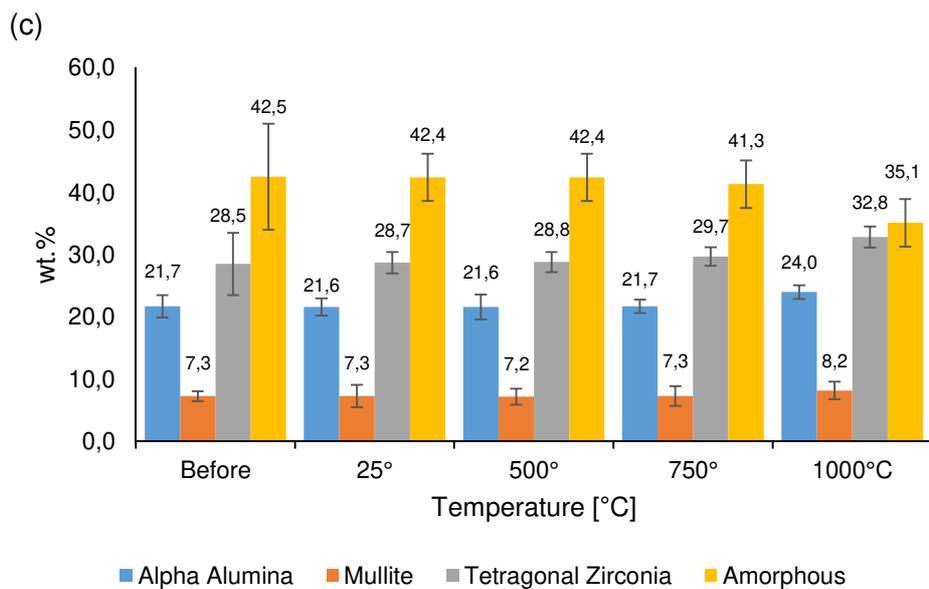
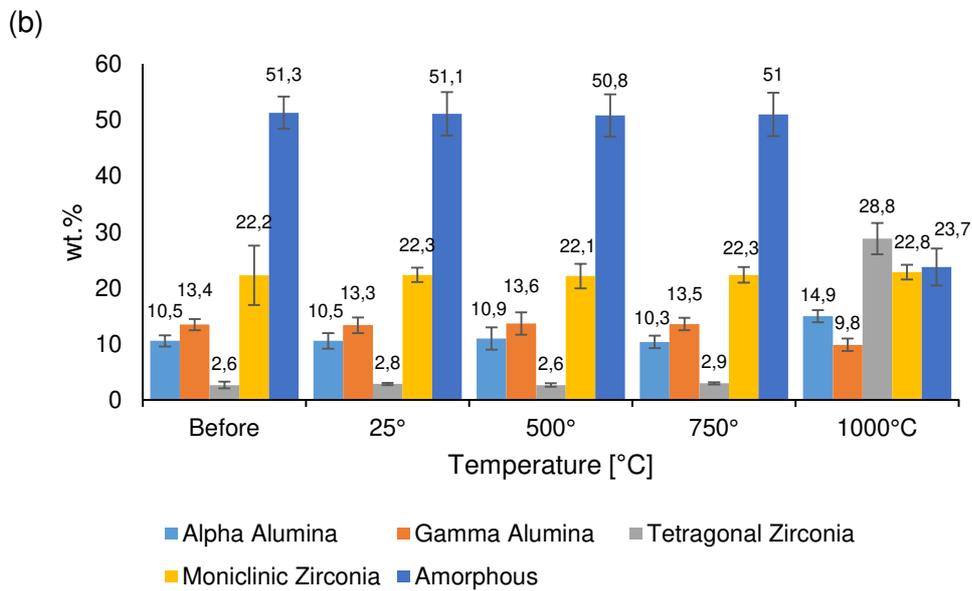
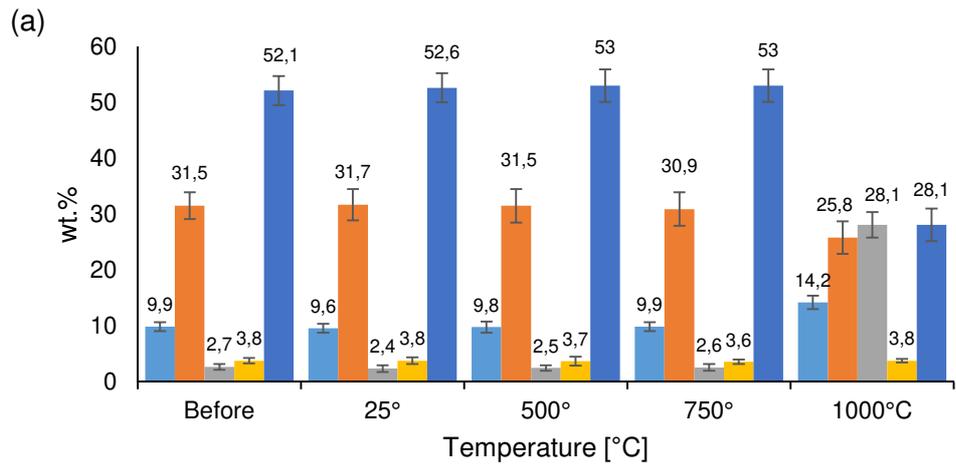


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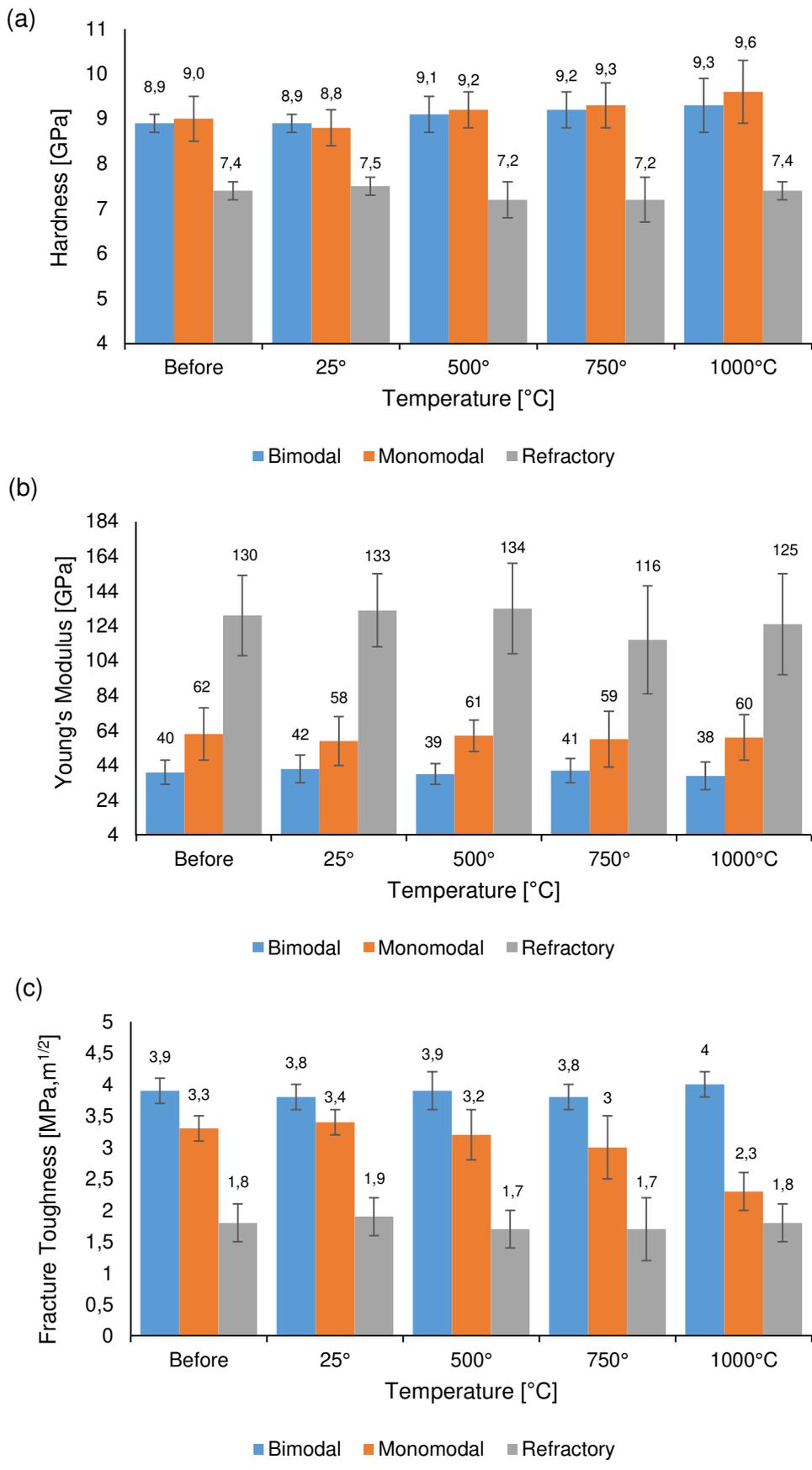


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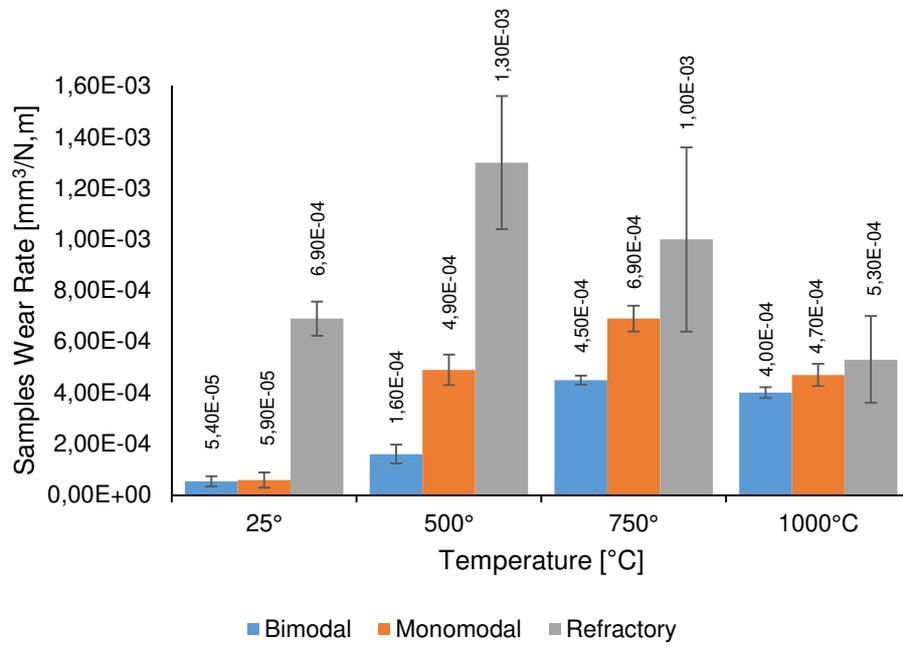


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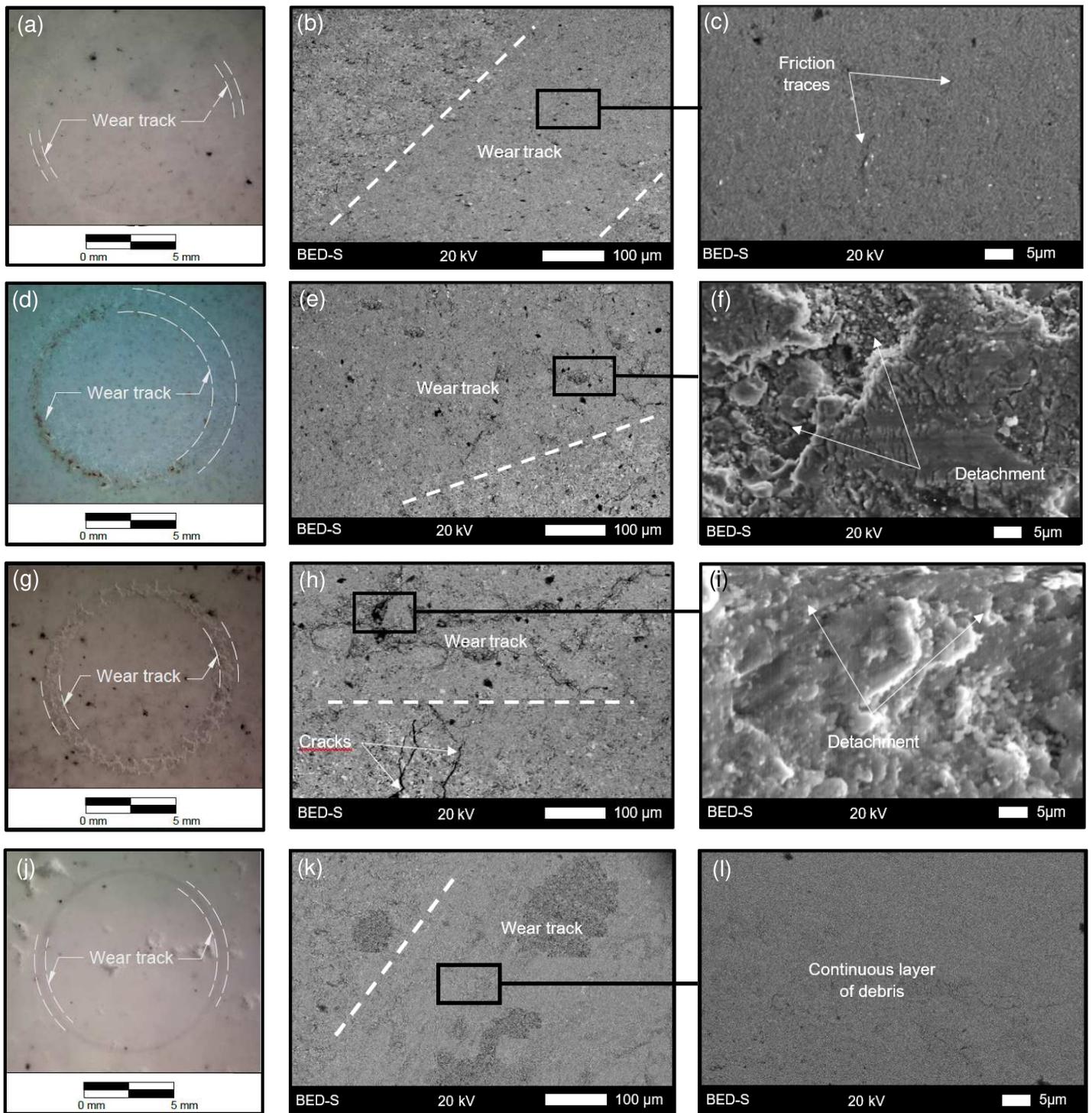


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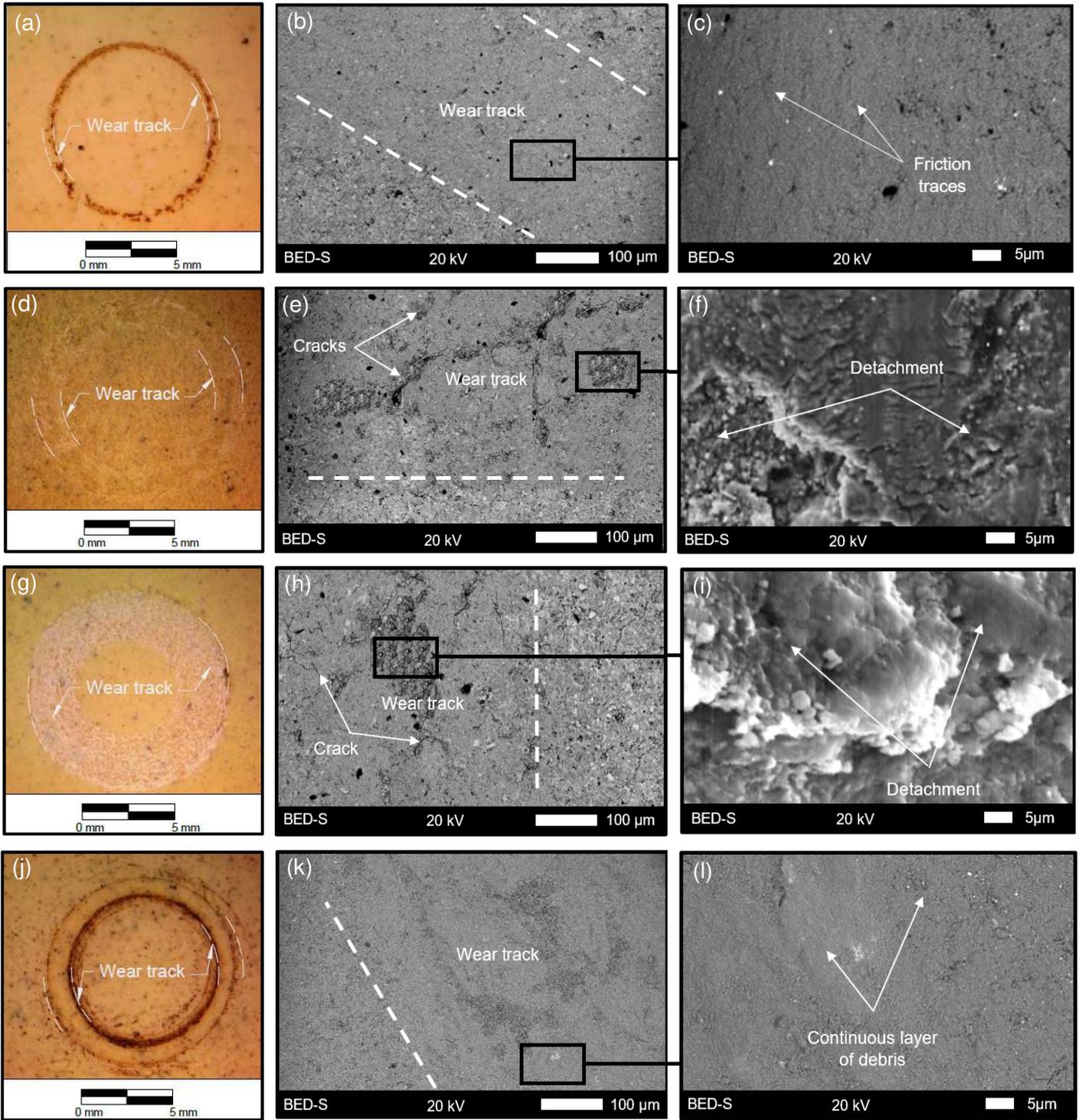


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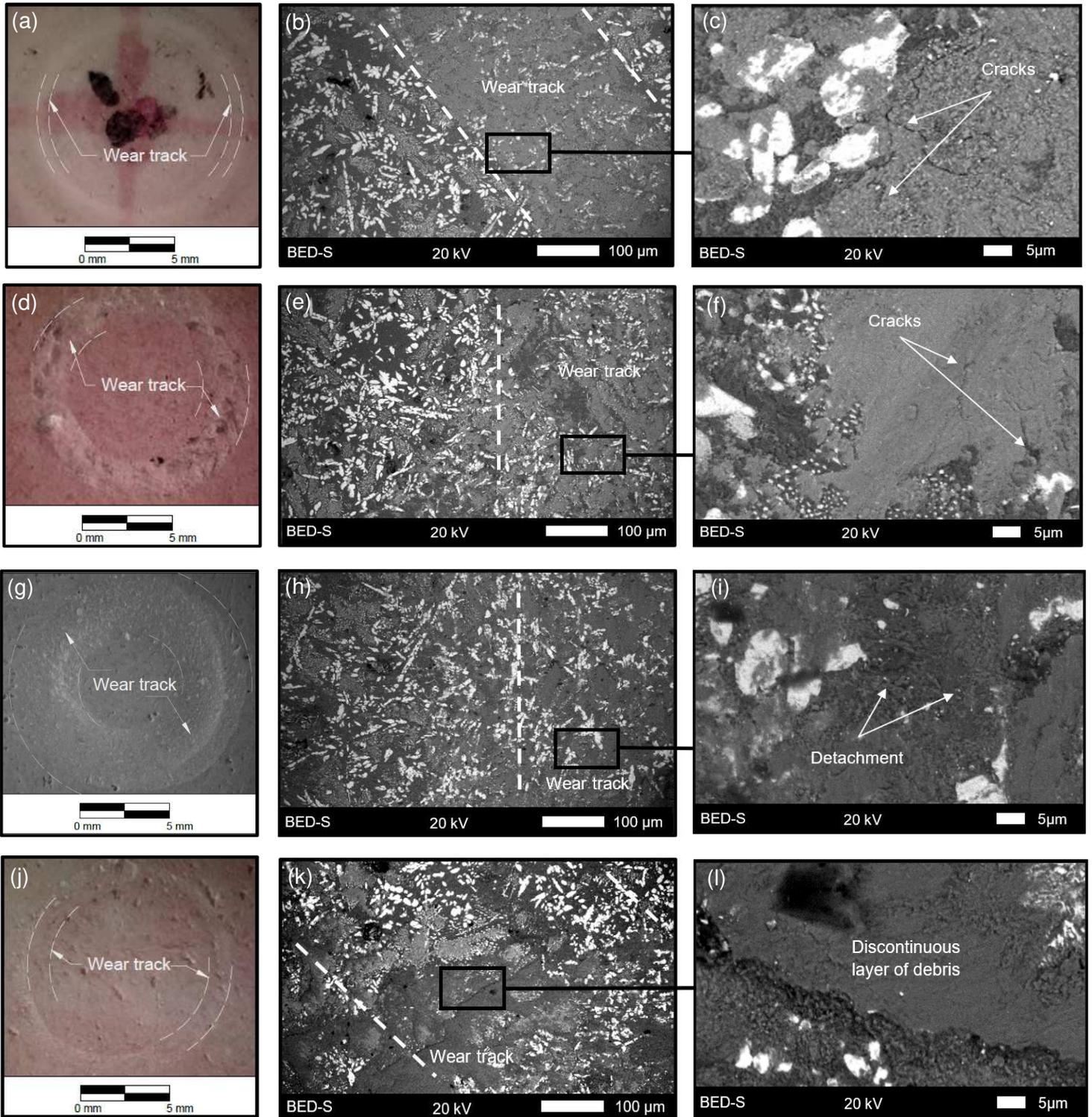


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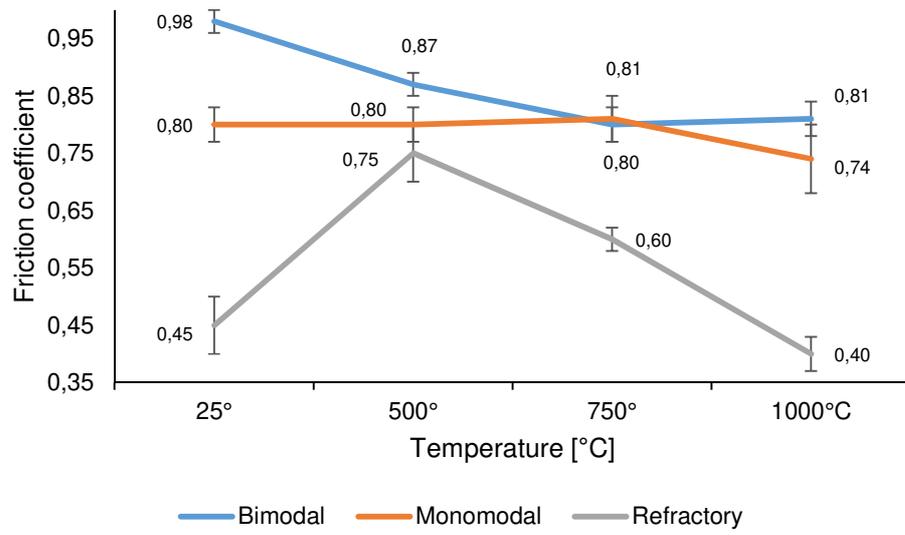


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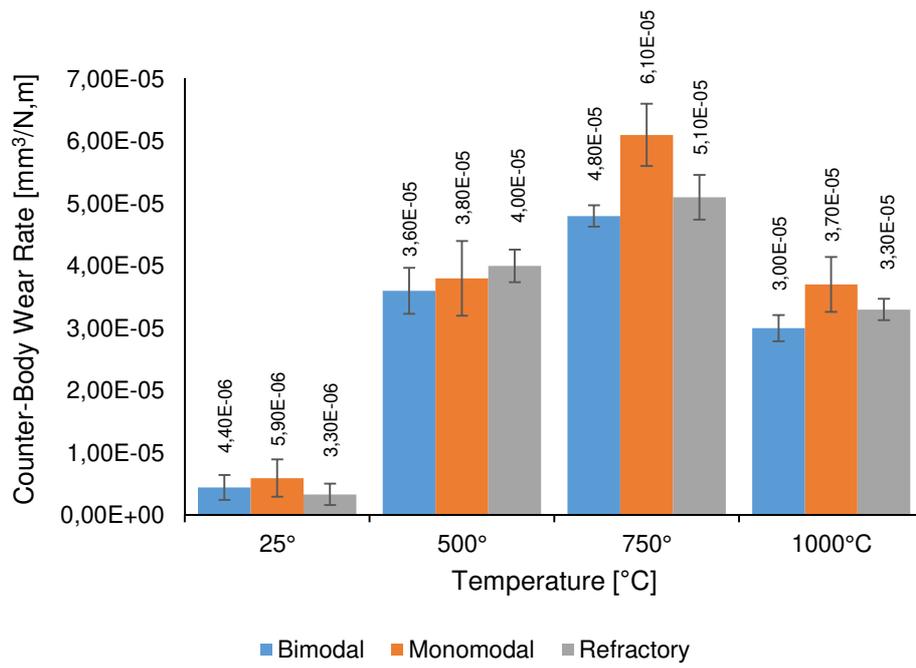


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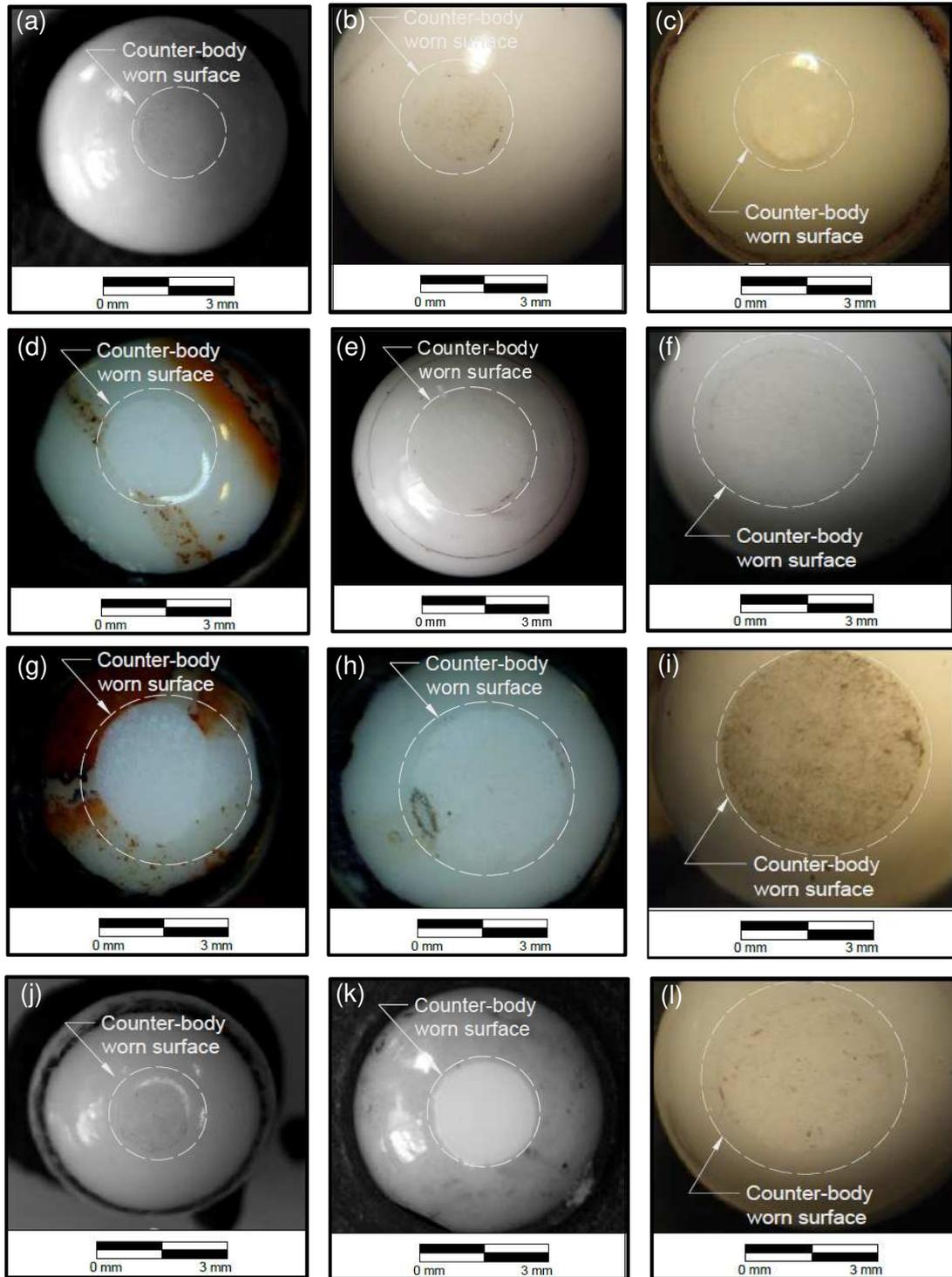


Figure 15.